"&ana¸ iva&ana AaiNa sausaMskar yaasaazl iSaxaNap`saar" iSaxaNamahiYa- D^a.baapUjal saaLuMKo

SHRI SWAMI VIVEKANAND SHIKSHAN SANSTHA'S, KOLHAPUR

Dattajirao Kadam Arts, Science & Commerce College, Ichalkaranji. DEPARTMENT OF CHEMISTRY

Academic year 2023-24

Syllabus: PREPARATION OF HOUSEHOLD CHEMICALS

Objectives:

- 1. To make availability of self employment.
- 2. To make confidence for doing job.
- 3. To develop skills required for job.
- 4. To develop skills of market survey.
- 5. To develop knowledge of household chemicals.

Course Duration: 30 Hrs Total: 100 Marks

Paper: I-Basic concept of Analytical chemistry Unit: I cleaning materials

[30 marks] [15 Marks]

1.1 Introduction of soaps

- 1.2 Theory of types
- 1.3 Raw materials
- 1.4 Types of soaps
- 1.5 Manufacture of soaps
- 1.6 Cleaning action of soaps
- 1.7 Introduction of soaps
- 1.8 Raw materials
- 1.9 Types of detergents, cationic, Anionic, Amphoteric and neutral
- 1.10 Preparation of Teepol and Deriphat
- 1.11 Comparison between Soaps

Unit: II Basic term in household chemicals

[10 marks]

- 2.1 Mole concept
- 2.2 Concentrations
- 2.3 Units of concentration
- 2.4 Normality
- 2.5 Molarity
- 2.6 Molality
- 2.7 Mole fraction
- 2.8 Percentage composition by volume
- 2.9 Percentage composition by weight

Unit: III Quantitative Analysis

[05 marks]

- 3.1 General introduction
- 3.2 Types of Titration
- 3.3 Indicators
- 3.4 Theory and types

Paper: II Household chemicals-introduction, synthesis and manufacturing [30 Marks] **Unit: I Household chemicals** [10 Marks] 1.1 Introduction of household chemicals or domestic chemicals 1.2 Sanitary acids 1.3 Phenyl 1.4 Liquid sops :Lysol, Dettol, Savlon 1.5 Acids-bases 1.6 General introduction 1.7 Theory 1.8 Types and strength of acids 1.9 Cleaning action of acid, handling and precaution 1.10Antiseptic and Disinfectant **Unit II: Phenols** [10 Marks] 2.1 Introduction of phenol 2.2 Methods of synthesis of phenols 2.3 Physical properties 2.4 Chemical properties 2.5 Uses Unit: III Basic of analytical techniques [10 Marks] 3.1 Purification of solids A) Crystallization:- preparation of saturation solution, hot Filtration, cooling of filtrate, filtration B) Distillation: simple distillation Paper-III:-Practical's [40 Marks] 1. Acid strength of Sanitary Acid 2. Preparation of Phenyl 3. Preparation of Soap

- 4. Preparation of Perfume
- 5. Preparation of Liquid Soap.
- 6. Preparation of Washing Powder
- 7. Preparation of Natural Indicator.
- 8. Saponification of oil.
- 9. Conductometric Titration I
- 10. Conductometric Titration -II
- 11. Preparation of Sanitizer