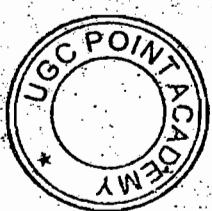
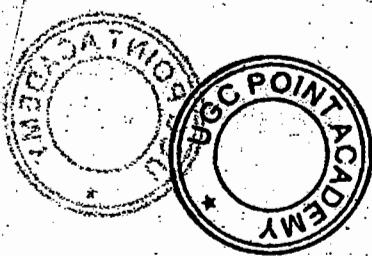


Jai Maa Saraswati

UGC-POINT ACADEMY
Statistical Mechanics



$$(a+b)^2 \rightarrow \frac{a^2 + 2ab + b^2}{\text{macro}} \quad \text{micro}$$

\rightarrow One coin: Two macrostate, Two microstate

\rightarrow 3 coin: 4-macrostate & 8-micro

Toss of n -coins;

$(a+b)^n \rightarrow (n+1) \rightarrow$ macrostate \rightarrow microstate.

Prob. of equally likely is $\frac{1}{2^n}$

→ If degeneracy of state are exist then no. of ways of distribution

$$R = \frac{LN}{A_{\text{eff}}} \cdot \prod_i g_i^{n_i}$$

Q. If 10 similar coins are tossed together, what is the prob. of getting

$$4 \text{ heads} = 10C_4 \left(\frac{1}{2}\right)^4 \left(\frac{1}{2}\right)^6$$

$$\rightarrow \frac{110}{1614} - \frac{1}{210}$$

Q. A random walker may take a step towards right or left at random, w/t is the prob. that the walker comes to its initial pos. after taking 6 steps $\Rightarrow 6C_3 (\frac{1}{2})^6$

Statistical Mechanics:

(Based on distribution of particle)

Branch of mechanics which deals with the system involving large no. of molecule/particles.

How?

→ S.M. solve the prob. of system by microscopic pt. of view.

→ In thermodynamics we solve the prob. of macroscopic system by macroscopic pt. of view.

→ By S.M. we can calculate the average properties of Bulk material. As the no. of molecules involve in the calculation increases the result of calculations becomes more perfect.

→ This mechanics is based on distribution of particle. By knowing the distribution of particles in system we calculate the physical properties of the system.

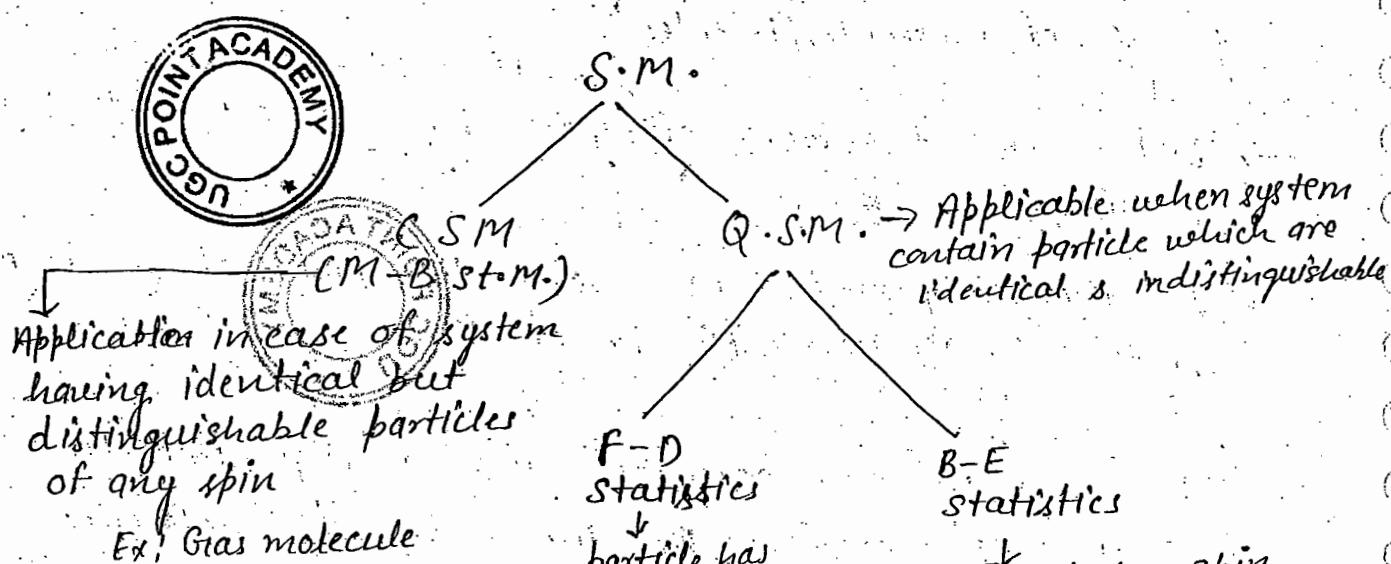
→ Initially this mechanics was developed by Maxwell, Boltzman, Gibbs etc.

→ This branch of S.M. is K/Las M-B statistical mechanics or classical statistical mechanics.
By it we can calculate the average energy (\bar{E}), Ave. (\bar{P}), Ave. temp (\bar{T}), Ave. value of chemical potential ($\bar{\mu}$) etc.

But laterly fails to explain BBK, specific heat, solids at low temp., BEC (Bose einstein condensin)

Then another branch of S.M. is developed by Fermi, Dirac, Bose, Einstein etc. which is based on Planck's quantum idea (discrete exchange of energy b/w radiation & matter).

→ This new branch is Klas. Q.S.M.



- Classical particle (A, B, C, etc) Name can be given.
- Quantum particles (..., ...) represented by dots of same size or same letter.

→ Total No. of particles (e^- , p , n)
 $= Z + A$ (atom)

A (nucleus)

Even → Bosons

Odd → fermions

${}_{\alpha}^3 Li^+$ is atom
Then Boson

${}_{\beta}^3 Li^+$ is nucleus
Then it is fermions.

π^- → fermions

An e^- is fermion.

→ A photon is boson,

C.S.M.:

Distribution:

Q. Prob: find the no. of ways of distribution, when 3 distinguishable particles are to be distributed in a box such that a) A box, contain 1 particle only $r=1$

Ans: 3

\boxed{A} or \boxed{B} or \boxed{C}

b.) Box contains two particles.

Ans: 3

\boxed{AB} , \boxed{BC} , \boxed{CA}

c.) Box contains 3 particles.

Ans: 1

\boxed{ABC}

→ Thus the no. of ways of distribution of N particle such that box contain r out of N distinguishable particle $= NCr = R$

Macrostate / or Maerascopic state:

The state which represents the observable property of the system, is called macrostate.

Observable properties: N (No. of particles), Volume (V)

Energy (E), momentum (p) etc.

→ A macrostate has large no. of microstates.
(Accessible under given restriction)

Microstate: The states which represents the unobservable property of the system.



→ No. of 4-state in a macrostate is called thermodynamical probability (\mathcal{P}). > 1. explain or represent no. of ways of distribution.

$$S = k_B \ln \Omega \rightarrow \text{As we increase no. of particles, more no. of microstates, so the system moves toward most prob. state.}$$

Prob:

N particles are to be distributed into two boxes such that box 1 contain n_1 particles and 2 box contain n_2 particles & $N = n_1 + n_2$. find the no. of ways of distributions? or thermodynamical probability/no. of 4-state

$$\Omega = N! / n_1! n_2!$$



$$\text{or } \frac{N!}{n_1! n_2!}$$

N particles are to be distributed in large No. of boxes having n_1, n_2, n_3, \dots particles

$$N = n_1 + n_2 + \dots$$

No. of ways of distribution:

$$= N! / n_1! n_2! n_3! \dots$$

$$\Omega = \frac{N!}{n_1! n_2! n_3! \dots} = \frac{N!}{\prod n_i!}$$

Prob: 1000 particles are to be distributed in 3 boxes having (500, 400 & 100) particles. find the entropy of the system.

$$\ln N = \ln(N!N - N) \quad \text{Stirling formula} \\ (\text{provided } N \text{ large})$$

$$\ln N = \ln(1, 2, 3 \dots N)$$

$$= \sum_{x_i=1}^N \ln x_i$$

$$= \int_{x=1}^N \ln x dx$$

$$= x \ln x - x \Big|_1^N$$

$$= N \ln N - N + 1$$



Prob: Two spin $\frac{1}{2}$ particles are placed in a mag. field in z-dir. find the total no. of ways of distributions.

U U
U D
D U
D D

$$\Omega = (2s+1)^N$$

En. of particle

En. of system

-HB -HB

2HB

-HB HB

0

HB -HB

0

D D 4B 4B

2HB

→ Total No. of 4-state = 4

→ The No. of 4-state corresponding to $E=0$

= 2

→ In all 4-state are equally probable. Find the probability of finding the system in most probable macrostate.

$$=\frac{1}{2}$$

Prob: Consider a system w/ 3 spin $1/2$ particles. The z-component of spin of system is given by

$$S_z = S_z(1) + S_z(2) + S_z(3)$$

each of $S_z(1), S_z(2), S_z(3)$ can take value $\pm 1/2$

find

- (i) Total no. of M-state of the system $\rightarrow 8$
- (ii) No. of M-states in macrostate $S_z = 1/2 \rightarrow 3$
- (iii) If all M-states are equally probable of finding the system in $S_z = -1/2 \rightarrow 318$

Prob: Consider the case of 10 unbiased coins, tossed simultaneously.

find the no. of ways of distribution

- i) 3 H uppermost (3H, 7T)

${}^{10}C_3$

Probability \rightarrow

The quantitative value of qualitative statements.

spin $1/2$ particles

Coin related prob.

Dice related "

Random walker

Particle distr. in two boxes

Baby birth

Particle in given volume V of a box of vol. V .

\rightarrow Prob. of equally likely probable even $= \frac{1}{2^n}$

Distribution of probability:

Binomial distribution: Is discrete distribution and applicable when p & $q (=1-p)$ are not too low. The prob. p of one trial is known.

$$(p+q)^N = 1 = {}^N C_0 p^0 q^N + {}^N C_1 p^1 q^{N-1} + \dots + {}^N C_r p^r q^{N-r} + {}^N C_N p^N q^0$$

Total Prob.
in n -trial.

with event

$$P(r, N-r) = {}^N C_r p^r q^{N-r}$$

unbiased



Prob: A coin is tossed 10 times. find the prob. of getting

$$(i) 1 H uppermost \quad r=1 \quad \frac{{}^{10} C_1 p^1 q^9}{p^1 q^9}$$

$$(ii) 9 T uppermost$$

$$(iii) At least 9 H uppermost \quad {}^{10} C_8 - {}^{10} C_9 \quad \text{HORROR}$$

$$(iv) 5 T uppermost \quad \text{(max. prob.)}$$

$$(v) (5T, 5H) distribution prob.$$

$$(vi) Prob. of least probable distribution \quad \frac{10!}{5!5!} \left(\frac{1}{2}\right)^5 \left(\frac{1}{2}\right)^5$$

$$(vii) Probability of most probable distri$$

$$(viii) Prob. At least 1T uppermost = {}^{10} C_0 \left(\frac{1}{2}\right)^0$$

$$P(1) + P(2) + \dots + P(10) = 1 - P(0)$$

$$p = \frac{1}{2}, \quad q = \frac{1}{2}, \quad N = 10$$

$$(ix) At most 1H uppermost \quad P(0) + P(1)$$

$$\Rightarrow {}^{10} C_0 \left(\frac{1}{2}\right)^0 \left(\frac{1}{2}\right)^9$$



$${}^{10} C_0 \left(\frac{1}{2}\right)^9 \left(\frac{1}{2}\right)^1 + {}^{10} C_1 \left(\frac{1}{2}\right)^1 \left(\frac{1}{2}\right)^9$$

$$\text{least} \Rightarrow {}^{10} C_0 \left(\frac{1}{2}\right)^0 \left(\frac{1}{2}\right)^0$$

Q. 10 unbiased coins are thrown simultaneously.

coin → baby
tail → brother
head → girl
tail → boy

coin → Dice

toss → Throw

p (prob. of coming a no. uppermost) $\{1, 2, 3, 4, 5, 6\}$



$$= \frac{1}{6}$$

$$q = 1 - p = \frac{5}{6}$$

Q. A dice is thrown 10 times. Find probability that

(i) no. 1 comes uppermost all times ${}^{10}C_{10} \left(\frac{1}{6}\right)^{10} \left(\frac{5}{6}\right)^0$

(ii) least probable case

(iii) No. 4 comes uppermost atleast 8 times.

$$P(8) + P(9) + P(10) = {}^{10}C_8 \left(\frac{1}{6}\right)^8 \left(\frac{5}{6}\right)^2 + {}^{10}C_9 \left(\frac{1}{6}\right)^9 \left(\frac{5}{6}\right)^1 + {}^{10}C_{10} \left(\frac{1}{6}\right)^0 \left(\frac{5}{6}\right)^{10}$$

(iv) No. 6 doesn't appear uppermost

$${}^{10}C_0 \left(\frac{1}{6}\right)^0 \left(\frac{5}{6}\right)^{10}$$

Q. 10 dice are thrown simultane-

Q. 10 balls are distributed into 2 boxes with equal prob.
find prob. of

(a.) all balls occupy 1st box $P(0, 10) = {}^{10}C_0 \left(\frac{1}{2}\right)^{10}$

(b.) Prob. of distribution $(8, 2) P(8, 2) = {}^{10}C_8 \left(\frac{1}{2}\right)^{10}$

(c.) Most probable distri.





Random Walker problem: One who don't care of taking steps

Child Drunkard

Prob: A one dim. random walker takes 10 steps in left-right with equal prob.

find prob. of

(i) Starting from origin, it will be at origin after all steps.

$${}^{10}C_5 \left(\frac{1}{2}\right)^{10}$$

(ii) Starting from 0, He will be one step away from 0, after all steps. Prob
 $n_1 + n_2 = 10$ { $n_1 = 5.5$
 Not happening $n_1 - n_2 = 1$

(iii) Most probable distribution ${}^{10}C_5 \left(\frac{1}{2}\right)^5 \left(\frac{1}{2}\right)^5$

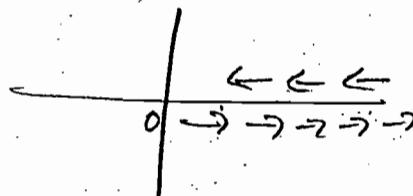
(iv) Least prob. distribution. ${}^{10}C_{10} \left(\frac{1}{2}\right)^{10}$

(v) Start from 0 will be 2 steps from 0 after all steps. $n_1 + n_2 = 10$, $(5, 5)$ or $(4, 6)$

$${}^{10}C_4 \frac{1}{2^{10}}$$

$$n_1 n_2 = 2$$

$$\underline{\text{Ans: } 2 \times {}^{10}C_6 \left(\frac{1}{2}\right)^{10}}$$



Prob: If 1-a random walker takes steps to right or left with equal prob. find the prob. that the random walker starting from the origin is back to origin after taking N even no. of steps. $Nc_{\frac{N}{2}} \left(\frac{1}{2}\right)^N = \frac{L^N}{L^{\frac{N}{2}} L^{\frac{N}{2}}} \left(\frac{1}{2}\right)^N$

Solⁿ:

$$= Nc_{\frac{N}{2}} \left(\frac{1}{2}\right)^N$$

$$= \frac{L^N}{\frac{N}{2}! \frac{N}{2}!} \left(\frac{1}{2}\right)^N$$



Prob: A drunkard start from a certain lamp-post on the street and he is trying to get his destination at a distance x . Some of steps he take in forward dirⁿ with prob. (p) & some steps in $\overleftarrow{\text{dir}}$ backward dirⁿ with prob. ($q = 1-p$), and each step is of equal length. Let P is the prob. that the drunkar will reach its destination after N -steps.

Solⁿ:

$$n_f + n_b = N$$

$$n_f L - n_b L = x$$

$$\text{Prob. } P(n_f) = Nc_{n_f} p^{n_f} q^{N-n}$$

$$= Nc_{n_f} p^{n_f} (1-p)^{n_b}$$

$$\rightarrow N = 10$$

$$L = 1m$$

$$x = 2m$$

$$p = \frac{2}{3}$$

$$q = \frac{1}{3}$$

$$n_f + n_b = 10$$

$$n_f - n_b = 2$$

$$n_f = 6$$

$$n_b = 4$$

$$= 10c_6 \left(\frac{2}{3}\right)^6 \left(\frac{1}{3}\right)^4$$

$n_f + n_b = N$
$n_f l_f - n_b l_b =$

$\rightarrow N = 10$ (number of step is constant)

$$l_f = 2m, l_b = 1m$$

$$x = 2m$$

$$P = \frac{2}{3}, Q = \frac{1}{3}$$

$$N_f + N_b = 10$$

$$2N_f - N_b = 2$$

$$N_f = 4$$

$$N_b = 6$$

$$= 10C_4 \left(\frac{2}{3}\right)^4 \left(\frac{1}{3}\right)^6$$



- Q. A random walker takes a step of unit length in +ve dirn with prob. $2/3$ & a step of unit length in the -ve dirn with prob. $1/3$. The mean displacement of walker after n step is

$$n/3, n/18, 2n/3, 0$$

Soln: Mean displacement

$$\langle x \rangle = \frac{P_0 x_0 + P_1 x_1 + P_2 x_2 + \dots + P_n x_n}{P_0 + P_1 + P_2 + \dots + P_n}$$

$$= 0 + nC_1 \left(\frac{2}{3}\right)^1 \left(\frac{1}{3}\right)^{n-1} \cdot 1 + nC_2 \left(\frac{2}{3}\right)^2 \left(\frac{1}{3}\right)^{n-2} \cdot 2 + \dots + nC_n \left(\frac{2}{3}\right)^n \left(\frac{1}{3}\right)^0 \cdot n$$

OR

$$\begin{aligned} \langle x \rangle &= \frac{P_f x_f - P_b x_b}{P_f + P_b} \\ &= \frac{\frac{2}{3}(n \times 1) - \frac{1}{3}(n \times 1)}{\frac{2}{3} + \frac{1}{3}} \end{aligned}$$

$$\langle x \rangle = \frac{n}{3}$$

Ave. distance travelled by the walker in forward dirn in n step with prob. $\frac{2}{3}$ of each step of length $1m$ unit

$$= \frac{2n}{3}$$

In Backward : mean
= $\frac{n}{3}$

$$\begin{aligned} \langle x \rangle &= P_x \\ &= \sum P_i x_i \end{aligned}$$

Step	Prob.
left	1/3
Right	2/3

→ Most prob

Q. A coin is tossed

Event Prob.

OH

1H

2H

3H

4H

5H

6H

7H

8H

9H

10H

Are Head

$$= OH \times P_0 + 1 \times P_1$$

+ ...



→ Statistical Error $\propto \frac{1}{\sqrt{N}}$
or
deviation

OR
Uncertainty $\propto \frac{1}{\sqrt{N}}$

→ Unbiased \rightarrow Biased coin

$$Ncr p^r q^{N-r}$$

$$P_H = \frac{1}{3}$$

$$2H = 2/3$$

$$P_T$$

$$H = 10$$

$$P(9H) = {}^{10}C_4 \left(\frac{1}{3}\right)^4 \left(\frac{2}{3}\right)^6$$

Q. Consider a box of volume v containing N identical but distinguishable particles. Find

a) The prob. that a given particle is in given volume v of the box.

$$\frac{Nv}{v}$$

$$\boxed{\frac{v}{V^N}}$$

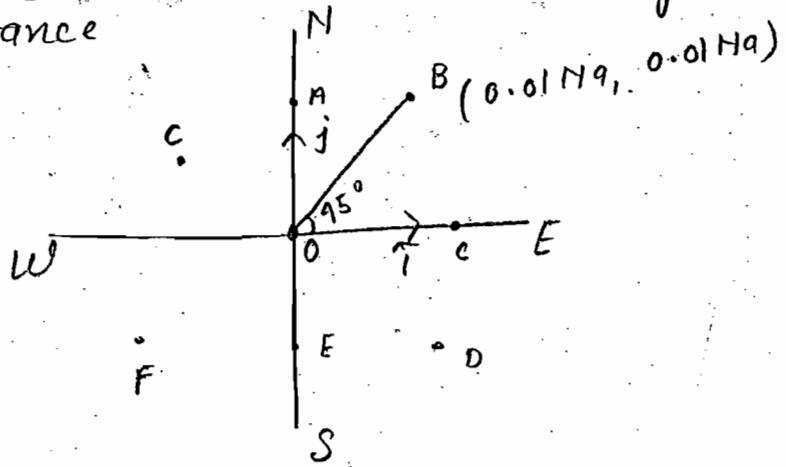
b) The prob. that a given set of n particles are in a given volume v & all other particles are in the remaining volume. $1, 2, 3, \dots, n(n+1) \dots N$

$$\left(\frac{v}{V}\right)^n \left(\frac{V-v}{V}\right)^{N-n}$$

c) Prob. of finding any n particles in volume v 's rest $(N-n)$ in remaining volume

$$= N_{Cn} \left(\frac{v}{V}\right)^n \left(1 - \frac{v}{V}\right)^{N-n}$$

Q. A child make a random walk on a square lattice of lattice constt a taking step in north, east, south & west dirⁿ with prob. 0.255, 0.255, 0.245, 0.245 respectively. After a large No. of steps, N , the expected position of the child w.r.t. the starting pt O is at a distance



$$\sqrt{2} \times 10^{-2} Na \text{ in } N-E \text{ dir}^n$$

$$\sqrt{2N} \times 10^{-2} a \text{ in } N-E \text{ dir}^n$$

$$\begin{aligned}
 \langle \vec{r} \rangle &= \frac{P_1 \vec{r}_1 + P_2 \vec{r}_2 + P_3 \vec{r}_3 + P_4 \vec{r}_4}{P_1 + P_2 + P_3 + P_4} \\
 &= 0.255 N \vec{a}^9 + 0.255 N \vec{a}^9 - 0.245 N \vec{a}^9 - 0.245 N \vec{a}^9 \\
 &= 0.01 N \vec{a}^9 + (0.015) N \vec{a}
 \end{aligned}$$

Prob:

spin $\frac{1}{2}$ particles are kept in a m.f. find prob.

(i) most probable distri

$$N_{C_{N/2}} \left(\frac{1}{2}\right)^N$$

(ii) $\left(\frac{2N}{3}, \frac{N}{3}\right)$ most prob. $N_{C_{2N/3}} \left(\frac{1}{2}\right)^N$

(iii) $(0, N)$ least prob

$$\rightarrow N_{C_0} \left(\frac{1}{2}\right)^N$$



Prob: Three fair dice are thrown simultaneously
find the probability of getting

a.) sum 18 uppermost

$$(6, 6, 6) \quad \frac{1}{216}$$

b.) sum of 12 uppermost

$$(6, 6) \quad \frac{1}{36}$$

c.) Most probable sum $\frac{m.p.}{7+0/12}$ as 8

$$(6, 1), (1, 6)$$

$$(3, 4), (4, 3)$$

$$(2, 5), (5, 2)$$

$$\begin{matrix} 6 & 2 \\ 2 & 6 \end{matrix}$$

$$\begin{matrix} 5 & 3 \\ 3 & 5 \end{matrix}$$

$$\begin{matrix} 4 & 4 \\ 4 & 4 \end{matrix}$$

$$\rightarrow \frac{1}{6}$$

$$\begin{matrix} 8fd \\ 13.6\% \end{matrix}$$

Prob: An unfair dice has prob. to appear 1, 2, 3, ..., 6 respectively as $\frac{1}{21}, \frac{2}{21}, \frac{3}{21}, \dots, \frac{6}{21}$. In throwing such a dice 4 times find prob.

(i) No. 1 appear in uppermost 2 times

$${}^4C_2 \left(\frac{1}{21}\right)^2 \left(\frac{20}{21}\right)^2 \quad N.C.P^N q^{4-n}$$

(ii) probable case



$${}^4C_4 \left(\frac{1}{21}\right)^4 \left(\frac{20}{21}\right)^0 \quad N.C.P^N q^0$$

$${}^4C_0 \left(\frac{6}{21}\right)^0 \left(\frac{15}{21}\right)^4$$

(iii) most probable case



(iv) If this dice is thrown 2 times. find the prob. of getting sum even uppermost

$$(1,1), (1,3), (1,5), (2,2), (2,4), (2,6)$$

$$(3,3), (4,4), (5,5), (6,6)$$

(v) In throwing such a dice 3 times find prob. of getting sum 17 uppermost $\rightarrow [1 - \text{odd sum}]$

$$(vi) \frac{1}{441} [1 + 6 + 10 + 4 + 16 + 48 + 9 + 46 + 30 + 25 + 36]$$

$$= \frac{225}{441}$$

$$(v) 1 - [(1,2)(1,4)(1,6)(3,2)(3,4)(3,6)(5,2)(5,4)(5,6)]$$

$$1 - \frac{2}{441} [2 + 4 + 6 + 6 + 12 + 18 + 10 + 20 + 30]$$

Q. A system has 3 particles having energy

$$E = -\epsilon_0 [J_1 J_2 + J_2 J_3 + J_3 J_1]$$

each of J_1, J_2, J_3 has values ± 1 . find

1.) min^m energy of the system $\rightarrow -3\epsilon_0$ (all -ve)

2.) max. en. of the system $\rightarrow \epsilon_0$ (Any 2 +ve or -ve) $3\epsilon_2$

3.) No. of H-state in microstate $E = \epsilon_0 = 6$ $3+3=6$

$$E = -3\epsilon_0 \text{ (macrostate)}$$

having 24 state

Total no. of H-state

$$= 8$$

Solⁿ:



Q. Consider a system having 3 parameter J_A, J_B & J_C each of J_A, J_B, J_C can take value ± 1 . The prob. of changing the value of these parameter is $2/3$ & not changing is $1/3$. Let $J_A = 1, J_B = 1, J_C = -1$ at a given instant. In the next instant the prob. that $J_A + J_B + J_C$ remain unchanged is $\frac{2}{3} \times \frac{1}{3} \times \frac{2}{3} \times \frac{4}{9}$ sum at instant

$$J_A + J_B + J_C = +1$$

given instant: 1 1 -1

on next, either

$$\frac{8}{27} + \frac{1}{9} \times \frac{2}{3} + \frac{1}{9} \times \frac{2}{3}$$

$$1 1 -1$$

$$-1 1 1$$

$$= \frac{10}{27}$$

when not change

$$\rightarrow \frac{1}{27} + \frac{4}{27} + \frac{4}{27} = \frac{9}{27}$$

Fundamental postulates of Statistical mechanics:

- Any gas under consideration may be considered to be composed of large no. of molecule which are constantly in motion. And behave like very small elastic sphere.
- The configuration of the particle is completely described by six co-ordinate (x, y, z, p_x, p_y, p_z). This six dim space is phase-space. This phase is divided into large no. of elementary cell called phase cell, each having equal size \hbar^3 .
- The most fundamental postulate of S.M. is "principal of equal a priori prob." ie all accessible macrostate are equally probable.
- The equilibrium state of gas corresponds to the macrostate of max^m prob.
- The total no. of molecule $N = \sum n_i = \text{constt}$ (conservation of mass)
- Total energy of the system remains constt.
 $E = n_1 E_1 + n_2 E_2 + \dots = \sum_i n_i E_i = \text{constt}$
 (energy conservation)

Prob: 10 particle are distributed in 3 given cells, with with energy per particle 0, ϵ , 3ϵ
 such that

$$n_1 = 3, n_2 = 5, n_3 = 2$$

If total E_{tot} of the system remains same.
 Then find $\delta n_1, \delta n_2$ when $\delta n_3 = -1$
 Also find the thermodynamical prob. for odd's new distribution.

Soln:

$$n_1 + n_2 + n_3 = 13 = \text{constt}$$

$$\delta n_1 + \delta n_2 + \delta n_3 = 0$$

$$\delta n_1 + \delta n_2 = 1$$

$$E = n_1 \epsilon_1 + n_2 \epsilon_2 + n_3 \epsilon_3 = \text{constt}$$

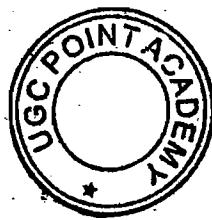
$$\Rightarrow E = n_2 \epsilon + 3n_2 \epsilon$$

$$0 = \epsilon \delta n_2 + 3 \epsilon \delta n_3$$

$$\delta n_2 = 3$$

new distribution (1, 8, 4)

No. of accessible 4-state / thermodynamic prob.



$$\Omega = \frac{L^N}{L^{n_1} L^{n_2} L^{n_3}} = \frac{L^{13}}{L^1 L^8 L^4}$$

$$\Omega_{\text{old}} = \frac{L^3}{L^3 L^5 L^5} \quad \begin{matrix} \text{most prob. distl.} \\ (\text{equilibrium state}) \\ (\text{max entropy state}) \end{matrix}$$

Phase-space: To describe the configuration of a particle completely, six co-ordinate (x, y, z, p_x, p_y, p_z) required.

Six dim. space is called phase space.

Volume element in phase space

$$\begin{aligned} d\Gamma &= dx dy dz dp_x dp_y dp_z \\ &= (dx dp_x) (dy dp_y) (dz dp_z) \end{aligned}$$

$$dx dp_x \geq h$$

$$dy dp_y \geq h^3$$

$$(dz dp_z) \geq h^3$$

Thus elementary cell in phase space or phase cell has volume

h^3

In cartesian co-ordinate

$(dt)_{\min} = 0 \rightarrow$ point element can never be possible in phase space.

\rightarrow Volume of phase space :

$$= \iiint dx dy dz db_x db_y db_z$$



$$-\infty < x, y, z < +\infty$$

$$-\infty < p_x, p_y, p_z < +\infty$$

$$\text{macrostates} \left\{ \begin{array}{l} p_x \rightarrow p_x + dp_x \\ p_y \rightarrow p_y + dp_y \\ p_z \rightarrow p_z + dp_z \end{array} \right.$$

$$\left\{ \begin{array}{l} p_x \rightarrow p_x + dp_x \\ p_y \rightarrow p_y + dp_y \\ p_z \rightarrow p_z + dp_z \end{array} \right.$$

$$p \rightarrow p + dp$$

$$E = E + dE$$

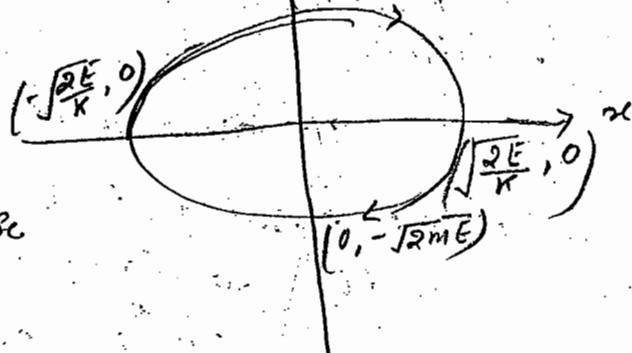


Application of phase space :

Ex: 1-D H.O.

$$E = \frac{p_x^2}{2m} + \frac{1}{2} K x^2$$

$$\Rightarrow \frac{p_x^2}{2mE} + \frac{x^2}{2E/K} = 1$$



Accessible area to the O.Sc

$$= \pi ab$$

$$= \pi \sqrt{2mE} \sqrt{\frac{2E}{K}}$$

$$= E \sqrt{2K} \sqrt{\frac{m}{K}} = \frac{E}{\nu}$$

No. of possible M-state :

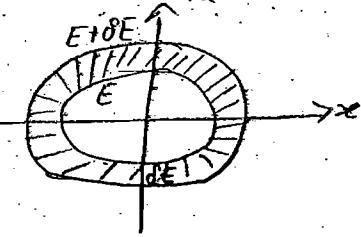
$$= \frac{\text{Accessible area to the O.Sc}}{\text{Area of 1 phase cell}}$$

$$n = \frac{E/\epsilon}{d\epsilon/dp_x} = \frac{E}{h\nu}$$

$$n = \frac{E}{h\nu}$$

$$E = nh\nu$$

$$n = 0, 1, 2, \dots$$



$$\delta n = \frac{\delta E}{h\nu}$$

$$\frac{\delta n}{\delta E} = \frac{1}{h\nu}$$

$$D(E) = \frac{1}{h\nu}$$

OR
 $D(E)\delta E = \frac{\delta E}{h\nu} = \delta n$

↓
 Density of state i.e. no.
 of states per unit
 energy range.

$$E \rightarrow E + \delta E$$

no. of states in en.
 range $\delta E = \frac{\delta E}{h\nu}$

$$\mathcal{N}(E)\delta E = \frac{\delta E}{h\nu}$$

$$0 \rightarrow E$$

$$\int_0^E \mathcal{N}(E)\delta E = \mathcal{N} = \text{Total no. of } n\text{-states in range } E$$

No. of n -state in momentum range $p \rightarrow p + dp$
 or

In energy range $E \rightarrow E + \delta E$

$$\mathcal{N}(p)dp = \frac{\text{Volume of phase space in mom. range}}{\text{Volume of 1 phase cell}}$$

$$\mathcal{N}(p)dp = \frac{\iiint dx dy dz \int_{p}^{p+dp} dp_x \int_{p_y}^{p_y + dp_y} dp_y \int_{p_z}^{p_z + dp_z} dp_z}{h^3}$$

$$= \iiint dx dy dz \int_{p}^{p+dp} \int_{p_y}^{p_y + dp_y} \int_{p_z}^{p_z + dp_z} dp_x dp_y dp_z$$

$$h^3$$

$$\begin{aligned} \mathcal{N}(p)dp &= \frac{V}{h^3} \int_p^{p+dp} \int_{p_y}^{p_y + dp_y} \int_{p_z}^{p_z + dp_z} p^2 \sin\theta d\theta d\phi dp \\ &\quad 0 < r < \infty \\ &\quad 0 < p < \infty \end{aligned}$$

$$= \frac{V}{h^3} 4\pi p^2 dp$$

$$\Omega(b)db = \frac{V}{h^3} \left[\frac{4}{3}\pi(p+db)^2 - \frac{4}{3}\pi p^2 \right]$$

$$\Omega(b)db = \frac{V 4\pi b^2 db}{h^3}$$

$$b = \sqrt{2mE}$$

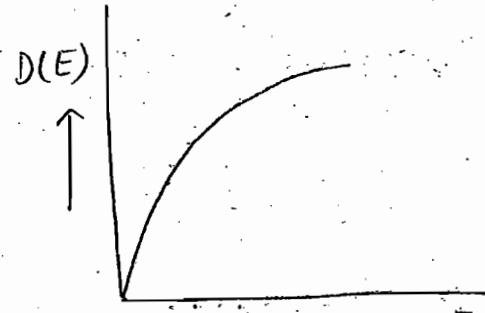
$$db = \sqrt{2m} \frac{1}{2} E^{-1/2} dE$$

$$dn = \Omega(E)dE = \frac{4\sqrt{2} V \pi m^{3/2} E^{1/2}}{h^3} dE$$

$$\Omega(E) = \frac{dn}{dE} = D(E) = \frac{4\sqrt{2} V \pi m^{3/2} E^{1/2}}{h^3}$$

Density
of state

$$D(E) \propto E^{1/2}$$



* Non-interacting photon gas:

No. of states available for photon is

mom. range $p \rightarrow p+dp$

or $E \rightarrow E+dE$

$$\Omega(b)db = \frac{V 4\pi b^2}{h^3} db$$

In terms of energy $b = E/c$

$$db = dE/c$$

$$dI(E)dE = \frac{V 4\pi E^2 dE}{c^3 h^3}$$

$$\frac{dI(E)}{dE} = \frac{V 4\pi E^2}{c^3 h^3}$$

II
D(E)

$$\frac{dN}{dE}$$

In terms of frequency

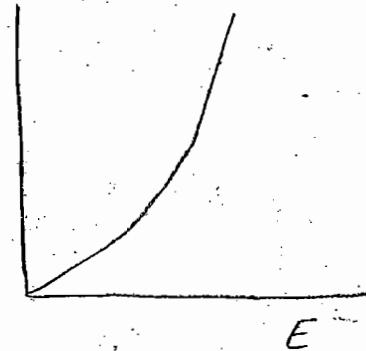
$$E = h\nu$$

$$dE = h d\nu$$

$$\boxed{\frac{D(\nu)}{D(\nu)} = \frac{V 4\pi \nu^2 d\nu}{c^3}}$$



$$D(E)$$



Phonon gas:

Expression are same as photongas.

* Density of states:

1-D: Consider a linear crystal of length \$L\$

The periodic boundary cond'n

$$\psi_K(x) = \psi_K(x+L) \quad x=0 \qquad x=L$$

$$A e^{ikx} = A e^{ik(x+L)}$$

$$e^{ikL} = 0 = e^{\pm i 2n\pi}$$

$$k = \pm \frac{2n\pi}{L} = -\frac{4\pi}{L}, -\frac{2\pi}{L}, 0, \frac{2\pi}{L}, \frac{4\pi}{L}$$

No. of k -state in range $k \rightarrow k + dk$

$$dn = \frac{\text{range of } k}{\text{length required for 1 state}}$$



$$= \frac{dk}{\left(\frac{2\pi}{L}\right)}$$

$$dn = \frac{L dk}{2\pi}$$

$$\frac{dn}{dk} = \frac{L}{2\pi}$$

OR

$D(k)$	$= \frac{L}{2\pi}$
DOS	



$$D(b) = \frac{L}{2\pi h}$$

from

$$D(k) dk = \frac{L}{2\pi} dk$$

$$D(b) db = \frac{L}{2\pi} \frac{db}{h}$$

$D(b) = \frac{L}{2\pi h}$

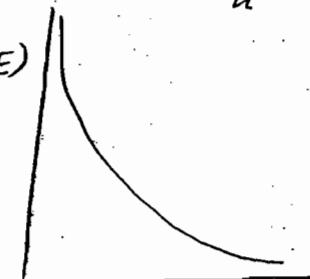
In terms of E

$$b = \sqrt{2mE} \Rightarrow db = \sqrt{\frac{m}{2}} E^{-1/2} dE$$

$$D(E) dE = \sqrt{2m} \frac{1}{2} E^{-1/2} dE = \sqrt{\frac{m}{2}} E^{-1/2} dE \frac{L}{h}$$

$D(E) dE = \frac{L}{h} \sqrt{\frac{m}{2}} E^{-1/2} dE$
--

$$D(E)$$



DOS in 2-D :

No. of μ -state in range $K \rightarrow K+dK$

$$dn = \frac{\text{Area available in phase space}}{\text{Area required for 1 state}}$$



$$\frac{2\pi K dK}{(\frac{2\pi}{L})^2}$$

$$dn = \frac{\pi K dK}{2\pi}$$

$$L^2 = A$$

$$D(K) dK = \frac{\pi K dK}{2\pi}$$

$$K = \frac{p}{\hbar}$$

$$dK = \frac{dp}{\hbar}$$

$$D(p) dp = \frac{\pi p dp}{2\pi \hbar^2}$$

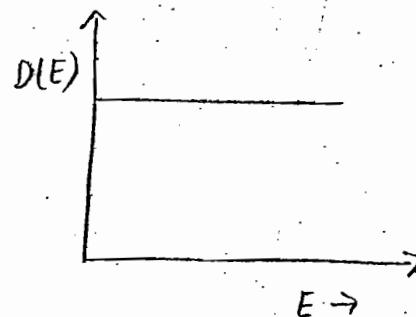
$$p^2 = 2mE$$

$$2pd\!p = 2m dE$$

$$pd\!p = m dE$$

$$D(E) dE = \frac{2\pi A m dE}{L^2}$$

$$D(E) \propto E^0$$



DOS in 3-D:

No. of 4-state in range

$$k \rightarrow k + dk$$

$$\text{or} \\ k_x \rightarrow k_x + dk_x, \quad k_y \rightarrow k_y + dk_y$$

$$k_z \rightarrow k_z + dk_z$$



$dn = \frac{\text{Volume available in phase space}}{\text{Volume required for 1 state}}$

$$dn = \frac{4\pi k^2 dk}{\left(\frac{2\pi}{L}\right)^3} = \frac{V k^2 dk}{2\pi^2}$$

$L^3 = V$

$$D(k) dk = \frac{V k^2 dk}{2\pi^2}$$

$$k = \frac{p}{\hbar}$$

$$dk = \frac{dp}{\hbar}$$

$$D(p) dp = \frac{V 4\pi p^2 dp}{\hbar^3} \rightarrow \text{Gr. E.}$$

$$p = \sqrt{2mE}$$

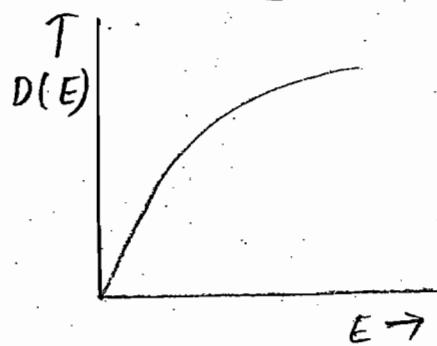
$$dp = \sqrt{\frac{m}{2}} E^{-1/2} dE$$

$$D(E) dE = \frac{4\sqrt{2} \pi m^{3/2} E^{1/2} dE}{\hbar^3}$$

$$D(E) \propto E^{1/2}$$

$$\rightarrow w = v_0 k$$

$$\frac{\hbar w}{v_0} = p = \frac{E}{v_0}$$



ask $\rightarrow E \propto p^2$ (given)

$s=2$ in case of particle
 $s=1$ for photon or phonon

$$D(E) \propto E^{q/s-1}$$

in d -dim

$\rightarrow E \propto p^2$ 3-dim

$$D(E) \propto E^{1/2}$$

$\rightarrow E \propto \sqrt{p}$ 3-D

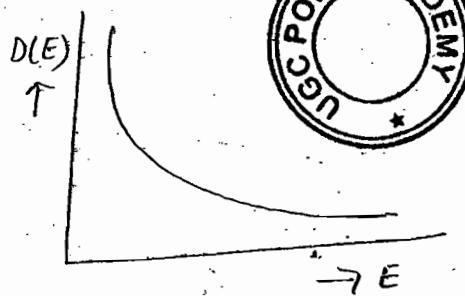
$$D(E) \propto E^{3/2-1} \propto E^5$$

Particle in 1-D rigid box \rightarrow

$$\begin{aligned} V(x) &= 0 & 0 < x < L \\ &= \infty & \text{elsewhere} \end{aligned}$$

$$D(E) \propto E^{-1/2}$$

$$\begin{aligned} E &= \frac{p^2}{2m} + 0 \\ D(E) &\propto E^{1/2-1} \end{aligned}$$

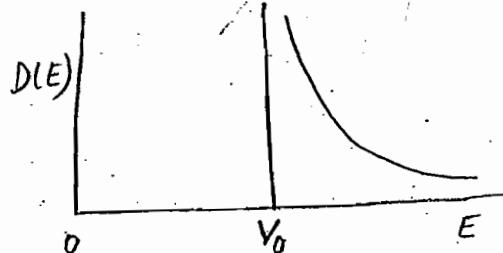


$$\rightarrow V(x) = V_0 \quad 0 < x < L$$

$$= \infty \quad \text{elsewhere}$$

$$E = \frac{p^2}{2m} + V_0$$

$$D(E) \propto (E - V_0)^{-1/2}$$



$$E \propto p^2$$

$$E \propto K^2$$

$$W \propto K^2$$

all are same things

$$\rightarrow D(E) = \frac{dn}{dE}$$

$$E = \frac{n^2 h^2}{8mL^2} + v_0$$

$$dE = \frac{n h^2 dn}{4mL^2}$$

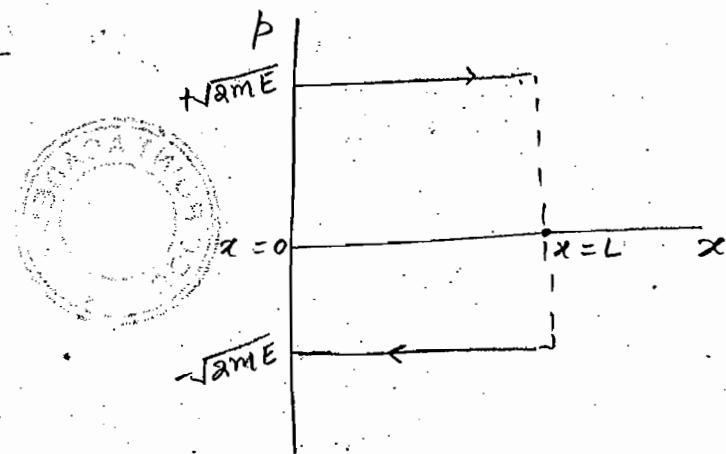
$$dn = \frac{4mL^2}{h^2} \cdot \frac{1}{n} \cdot dE$$

$$\times \frac{1}{n} \propto \frac{1}{(E - v_0)^{1/2}}$$

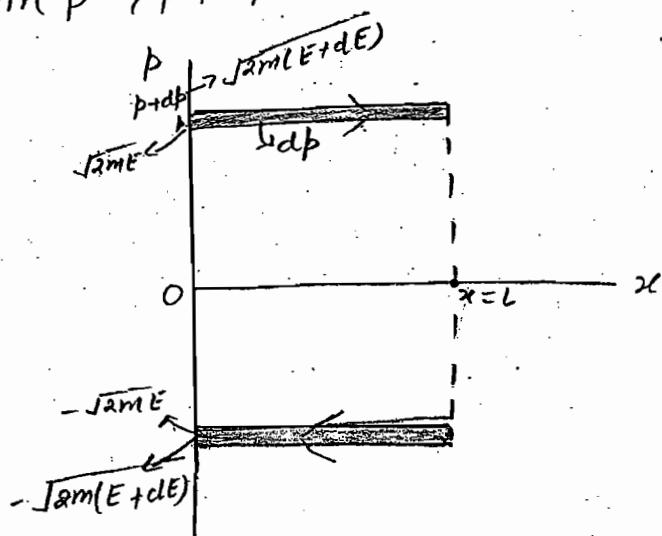
Phase-space diagram:

$$E = \frac{p^2}{2m}, 0 < x < L$$

$$p = \pm \sqrt{2mE}$$



→ If p -varies from $p \rightarrow p + dp$



Prob: The Hamiltonian of a particle is given by

$$H = \frac{p^2}{2m} - \frac{\alpha z^2}{2}$$

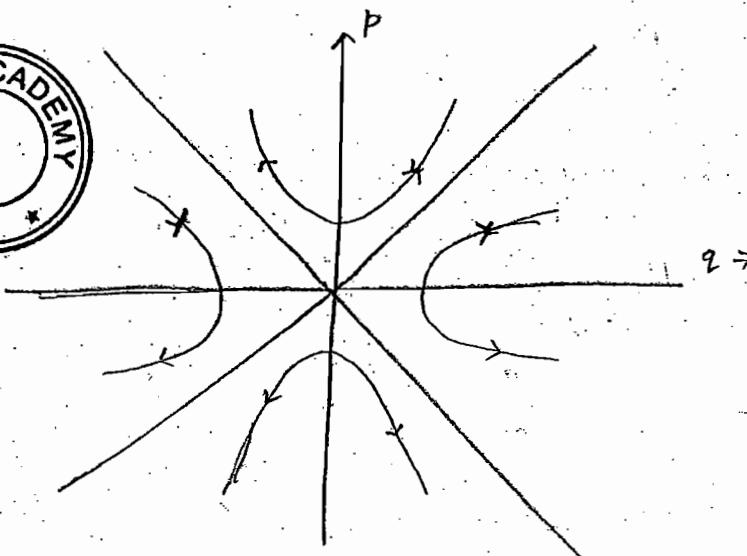
which of the following fig. represents the motion of the particle in phase space.

$$H = \frac{p^2}{2m} - \frac{\alpha z^2}{2}$$

$E \rightarrow \text{constt.}$

P.S.-digi: Parabola

$$V(x) = -\frac{\alpha z^2}{2} \text{ i.e. } \text{eqn of para}$$



$$\dot{q} = \frac{\partial H}{\partial p} = \dot{p}_z$$

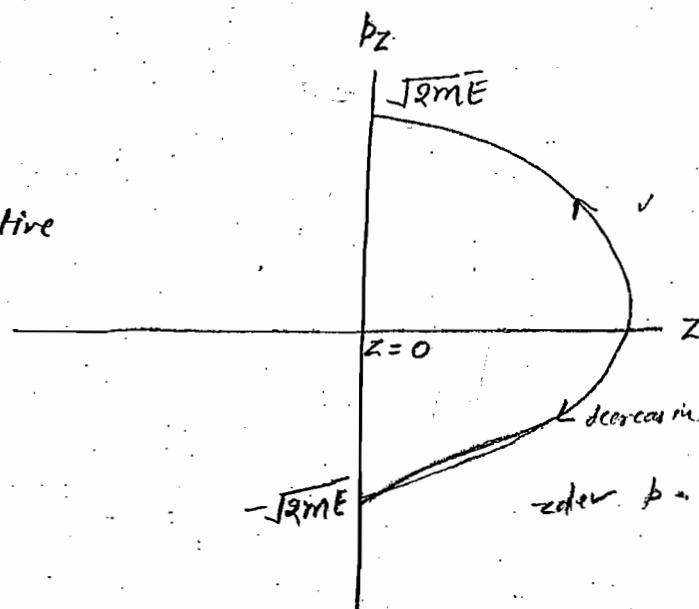
$$\dot{p} = -\frac{\partial H}{\partial q} = \alpha q$$

- * The trajectory (phase space diagram) on $z-p_z$ plan of ball bouncing from a perfect elastic hard sphere.

$$H = \frac{p_z^2}{2m} + mgz$$

$E = \text{constt}$

system is conservative



→ If the particle is moving from inelastic sphere :

Elastic

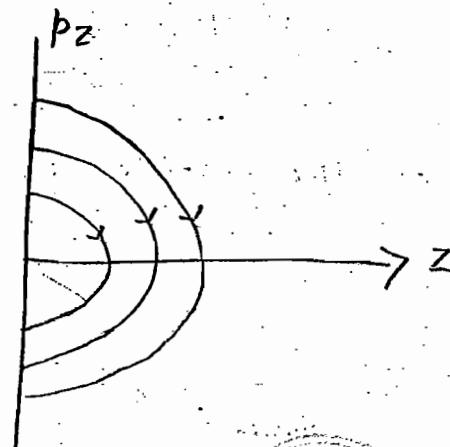
$$\vec{P} = \vec{P}' \quad T = T'$$

$$\vec{P} \neq \vec{P}'$$

$\vec{P} \neq \vec{P}'$ Inelastic

$$\vec{P} \neq \vec{P}'$$

$$T = T'$$



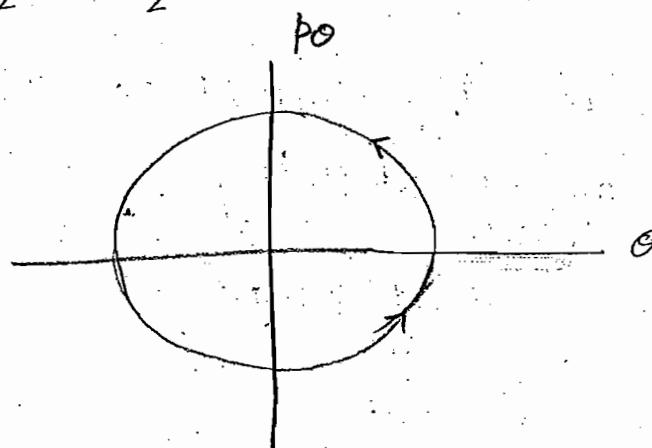
→ Phase space diagram of simple pendulum :

$$H = \frac{p_\theta^2}{2ml^2} + mgl(1 - \cos\theta)$$

θ is small

$$\cos\theta = 1 - \theta^2/2$$

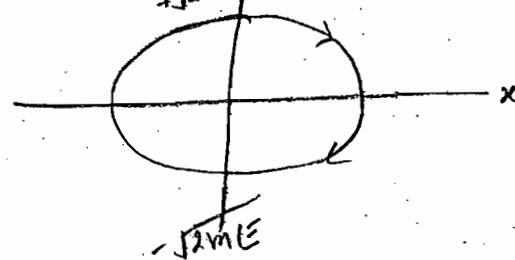
$$H = \frac{p_\theta^2}{2ml^2} + \frac{mgl\theta^2}{2}$$



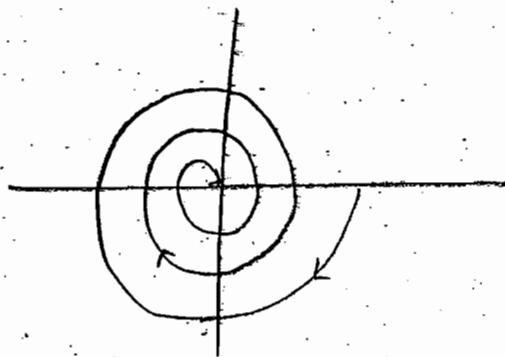
$$E = \frac{p_x^2}{2m} + \frac{1}{2}Kx^2$$

$$p_x = -Kx \quad \Rightarrow \quad \frac{dp_x}{dt} = -Kx$$

+ J.m.t



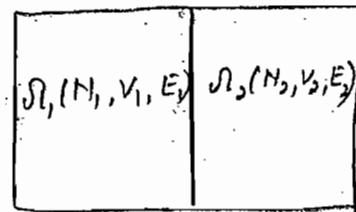
→ If bob is immersed in water then phase space diagram



* Boltzmann's Definition of Entropy →

$$S = k_B \ln \Omega$$

Consider two independent systems having macrostates are described by (N_1, V_1, E_1) & (N_2, V_2, E_2) with thermodynamic probabilities $\Omega_1(N_1, V_1, E_1)$ & $\Omega_2(N_2, V_2, E_2)$ respectively.



Thermodynamical probability of composition

$$\Omega = \Omega_1 \times \Omega_2 \quad \text{①}, \quad \text{En. of composition } E = E_1 + E_2 \quad \text{②}$$

If energy is allowed to exchange

$$\Omega = \Omega_1(E_1) \times \Omega(E_2)$$

Let equilibrium energy is \bar{E}

$$\Omega = \max$$

$$\left. \frac{\partial \Omega}{\partial E_1} \right|_{E=\bar{E}} = 0$$

$$0 = \Omega_2 \frac{\partial \Omega_1}{\partial E_1} \left. \frac{\partial E_1}{\partial E_1} \right|^{+2} + \left. \frac{\partial \Omega_2}{\partial E_2} \frac{\partial E_2}{\partial E_1} \right|^{-1} \Omega_1$$

$$\frac{1}{\Omega_1} \frac{\partial \Omega}{\partial E_1} = \frac{1}{\Omega_2} \frac{\partial \Omega_2}{\partial E_2}$$

$$\frac{1}{\mathcal{N}} \cdot \frac{\partial \mathcal{N}}{\partial E} = \text{constt} \quad -③$$

$$\frac{\partial}{\partial E} (\ln \mathcal{N}) = \text{constt} \quad -④$$

$B = \frac{1}{k_B T}$ so entropy $= \frac{1}{T}$

By thermodynamic

$$\left. \frac{\partial S}{\partial E} \right|_{N,V} = \frac{1}{T} \quad -④$$

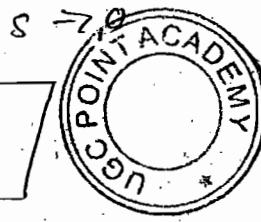
from
 $TdS = dE + PdV - \mu dN$

Dividing ⑤ by ④

$$\frac{\partial S}{\partial (\ln \mathcal{N})} = \frac{1}{B T} = k_B$$

$$S = k_B \ln \mathcal{N} + C$$

$$C = 0 \quad \text{at } T \rightarrow 0K$$



$$S = k_B \ln \mathcal{N}$$

$$\frac{\partial S}{\partial V} = \frac{P}{T}$$

→ If we allow volume to exchange

$$\frac{\partial}{\partial V} (\ln \mathcal{N}) = \eta = \frac{P}{k_B T} = \frac{P}{T}$$

If allow N to exchange

$$\frac{\partial}{\partial N} (\ln \mathcal{N}) = \gamma = -\frac{1}{k_B T}$$

$$\left. \frac{\partial \Omega}{\partial E} \right|_{N,V} = \frac{1}{T}$$

$$\left. \frac{\partial S}{\partial V} \right|_{N,E} = \frac{P}{T}$$

$\Omega \rightarrow$ statistical entropy

$T \rightarrow$ statistical temp.

$\beta = \text{inverse temp.}$

Ph. E. $S = k_B \Omega \rightarrow S(E)$

$$\left. \frac{\partial \Omega}{\partial E} \right|_{N,V} = \frac{1}{T}$$

$$\left. \frac{\partial \Omega}{\partial V} \right|_{E,N} = \frac{P}{T}$$

$$\left. \frac{\partial \Omega}{\partial N} \right|_{E,V} = -\frac{1}{T}$$

Prob: Consider a system of N magnetic atoms with spin $\frac{1}{2}$. At low temp. the system is ferromagnetic while at high temp. T it is paramagnetic. Neglecting all DOF except spins, find entropy of the system at.

(i) $T \rightarrow 0$ (ii) $T \rightarrow \infty$

(i) $T \rightarrow 0$

Distribution of spins

→ Almost all spins are aligned in same dirⁿ.

No. of ways of distribution

$$\Omega \rightarrow 1$$

$$S = k_B \ln \Omega$$

$$S \rightarrow 0$$

grif
(ii)

$T \rightarrow \infty$

System attain equilibrium.

$$\Omega = \text{max.}$$

$$\begin{aligned} \text{No. of distr.} &= (2s+1)^N = \underbrace{(2 \times \frac{1}{2} + 1)}_2^N = 2^N \\ \text{Max. no. of microstate} & \end{aligned}$$

$$\text{Entropy } S = k_B \ln(2^N)$$

$$= (\ln 2^N k_B)$$

Q. Consider a system of N paramagnetic atoms each have mag. moment M , are placed in mag. field B . n atoms are aligned $\parallel B$ & $(N-n)$ aligned antile to B . Find

(i) Internal En. of the system

(ii) Entropy of the system

(iii) The thermodynamic temp. of the system.

$$(i) E = nE_{\parallel B} + (N-n)E_{\text{antile}}$$

$$= n(-HB) + (N-n)HB$$

$$E = (N-2n)HB$$

$$(ii) \quad \sigma = \frac{dn}{n} = \frac{1N}{\ln \frac{1N}{N-n}}$$

$$S = k_B \ln \sigma$$

$$= k_B \ln \left(\frac{1N}{\ln \frac{1N}{N-n}} \right)$$

$$= N k_B \ln N - N k_B - N k_B \ln n + n k_B \\ - (N-n) k_B \ln (N-n) + (N-n) k_B$$

$$S = k_B \left[N \ln \left(\frac{N}{N-n} \right) - n \ln \left(\frac{n}{N-n} \right) \right]$$

$$(iii) \quad \frac{1}{T} = \frac{\partial S}{\partial E} = \frac{\partial S}{\partial E} \cdot \frac{\partial E}{\partial E}$$

$$= \frac{\partial S}{\partial n} \left(-\frac{1}{2k_B} \right)$$

$$= \frac{k_B}{2k_B} \ln \left(\frac{n}{N-n} \right)$$

$$\frac{1}{T} = \frac{2k_B}{k_B \ln \left(\frac{n}{N-n} \right)}$$

$$T = \frac{2k_B}{\ln \left(\frac{n}{N-n} \right)}$$

Distribution Laws :-

Boltzmann's distribution law:

→ Applicable to classical particles (Identical, distinguishable or any spin.)

Consider a system of N particles system has different energy states having energies $\epsilon_1, \epsilon_2, \dots$.
The no. of ways of distributions of particles such that :

ϵ_1 energy state has n_1 particles.

ϵ_2 energy state has n_2 particles.

$$= N c_{n_1} \times {}^{N-n_1} c_{n_2} \times \dots$$

$$= \frac{LN}{L n_1 L n_2 L n_3}$$

$$= \frac{LN}{\prod_i^n n_i}$$

when states are
non-degenerate

If energy state are degenerate

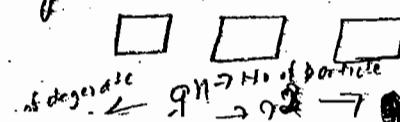
i.e., ϵ_1 energy state has degeneracy g_1
 ϵ_2 " " " " g_2

No. of ways of distribution

$$\mathcal{N} = \frac{LN}{\prod_i^n n_i} g_1^{n_1} \times g_2^{n_2} \times \dots$$

~~$\frac{LN}{\prod_i^n g_i^{n_i}}$~~ — 1

ϵ_1 : En. states has two particles A, B
If deg. of this state is 3.



q. $N = 3$
 No. of state ϵ_1, ϵ_2
 $g_1 = 3 \quad g_2 = 1$
 $n_1 = 2 \quad n_2 = 1$

find No. of ways of distribution

→ At equilibrium

$$\mathcal{N} = \text{max.} = \ln \Omega$$

with constraints



$$N - \sum_i n_i = 0$$

$$E - \sum_i n_i \epsilon_i = 0$$

$$\frac{\partial f}{\partial x_i} + \sum_K \lambda_K \frac{\partial \phi_K}{\partial x_i} = 0$$

$$\frac{\partial \Omega}{\partial n_i} + \alpha \frac{\partial \phi_1}{\partial n_i} + \beta \frac{\partial \phi_2}{\partial n_i} = 0$$

$$\frac{\ln g_i}{n_i} - \alpha - \beta \epsilon_i = 0$$

OR

$$\frac{\partial (\ln \Omega)}{\partial n_i} + \alpha \frac{\partial \phi_1}{\partial n_i} + \beta \frac{\partial \phi_2}{\partial n_i} = 0$$

from eqⁿ ①

$$\ln \Omega = \ln N - \sum_i n_i \ln n_i + n_i \sum_i \ln g_i$$

$$\frac{\partial (\ln \Omega)}{\partial n_i} = - \ln n_i - \frac{n_i}{n_i} + 1 + \ln g_i$$

$$\frac{\ln g_i}{n_i} = \alpha + \beta \epsilon_i$$

$$\frac{g_i}{n_i} = e^{\alpha + \beta \epsilon_i}$$

$$n_i = \frac{g_i}{e^{\alpha + \beta \epsilon_i}}$$

$$f(x, y, z) = \text{max}$$

$$\phi_1(x, y, z) = 0$$

$$\phi_2(x, y, z) = 0$$

$$\frac{\partial f}{\partial x} + \sum_K \lambda_K \frac{\partial \phi_K}{\partial x} = 0$$

$\lambda_K \rightarrow$ undetermined multiplier

$$df = \frac{\partial f}{\partial x} dx + \frac{\partial f}{\partial y} dy + \frac{\partial f}{\partial z} dz$$

$$= 0$$

$$d\phi_1 = \frac{\partial \phi_1}{\partial x} dx + \frac{\partial \phi_1}{\partial y} dy$$

+ ...

$$d\phi_2 = \frac{\partial \phi_2}{\partial x} dx + \dots$$

$$df + \lambda_1 d\phi_1 + \lambda_2 d\phi_2 = 0$$

α_i, β are undetermined constants.

$$n_i \propto g_i$$

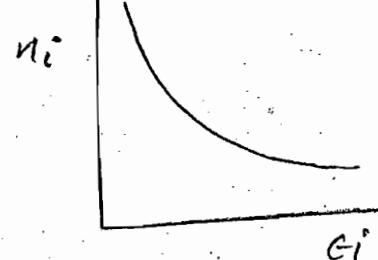
g_i = priori probability

or
Degeneracy of the E_i energy state

$$\rightarrow n_i = g_i e^{-\alpha} e^{-\beta E_i}$$

$$= A g_i e^{-\beta E_i}$$

$$A = e^{-\alpha} = \text{constt}$$



$$\sum n_i = N = \sum_i A g_i e^{-\beta E_i}$$

$$A = \frac{N}{\sum_i g_i e^{-\beta E_i}}$$



$$n_i = \frac{N g_i e^{-\alpha} e^{-\beta E_i}}{\sum g_i e^{-\alpha} e^{-\beta E_i}}$$

Probability:

$$P(E_i) = \frac{n_i}{N} = \frac{g_i e^{-\beta E_i}}{\sum_i g_i e^{-\beta E_i}}$$

$$Z = \sum g_i e^{-\beta E_i}$$

↓
Partition function

→ When states are non-degenerate i.e. $g_i = 1$

$$Z = \sum_i e^{-\beta E_i} = \text{sum of all Boltzmann's factor}$$

Prob: calculate the partition fun in case of 1D H.O.

$$E_n = \left(n + \frac{1}{2}\right) \hbar\omega \quad Q, osc$$

$$E = \frac{\hbar^2}{2m} + \frac{1}{2} m\omega^2 x^2 \quad cl. osc$$

$$Z = \sum_n g_n e^{-\beta E_n}$$



$$\begin{aligned} Z &= \sum_n e^{-\beta E_n} \\ &= \sum_{n=0}^{\infty} e^{-\beta \left(n + \frac{1}{2}\right) \hbar\omega} \end{aligned}$$

$$Z = \frac{a}{1 - e^{-\beta \hbar\omega/2}} + e^{-3\beta \hbar\omega/2} + \dots$$

$$\frac{a}{1 - r}$$

$$= \frac{e^{-\beta \hbar\omega/2}}{1 - e^{-\beta \hbar\omega}}$$

Partition fun^q for 1D cl. osc:

$$E = \frac{\hbar^2}{2m} + \frac{1}{2} m\omega^2 x^2$$

$$Z = \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \frac{dx d\beta x}{h} e^{-\beta E}$$

rigid rot.

$$E_J = (J^2 + J) \frac{\hbar^2}{2I}$$
$$= \int_0^\infty (2J+1) e^{-\beta E_J} dJ$$



Q. If two ideal dice are rolled once, what is the probability of getting at least one 6.

$$\frac{11}{36}, \frac{1}{36}, \frac{10}{36}, \frac{5}{36}$$

Partition function:

$$\begin{aligned}
 Z &= \sum q_i e^{-\beta E_i} \\
 &= \sum e^{-\beta E_i} \\
 &= \frac{1}{h^3} \int e^{-\beta E_i} dx dp_x \\
 &= \frac{1}{h^3} \iiint e^{-\beta E} dq_1 dq_2 dq_3 dp_1 dp_2 dp_3 \\
 &= \text{tr}(e^{-\beta \hat{H}})
 \end{aligned}$$

1-D Q.H.O.

$$E_n = \left(\frac{n+1}{2}\right) \hbar \omega$$

non-degenerate case $\Rightarrow q_n = 1$

$$Z = \frac{e^{-\beta \hbar \omega / 2}}{1 - e^{-\beta \hbar \omega}}$$

1-D Classical oscillator:

$$Z = \frac{1}{\beta \hbar \omega} = \frac{k_B T}{\hbar \omega}$$

Partition funⁿ of rigid rotator:

$$Z = \frac{2 I k_B T}{\hbar^2}$$

Partition funⁿ for a molecule of classical ideal gas-

$$E = \frac{p^2}{2m} = \frac{p_x^2}{2m} + \frac{p_y^2}{2m} + \frac{p_z^2}{2m}$$

~~$$Z = \frac{1}{h^3}$$~~

$$Z = \int q(p) dp e^{-\beta E}$$

No. of macro state or No of non-degenerate state

$$Z = \iiint \frac{dx dy dz dp_x dp_y dp_z}{h^3} e^{-\beta \left(\frac{p_x^2}{2m} + \frac{p_y^2}{2m} + \frac{p_z^2}{2m} \right)}$$



$$= \frac{V}{h^3} \int_{-\infty}^{\infty} e^{-\beta p_x^2/2m} dp_x \int_{-\infty}^{\infty} e^{-\beta p_y^2/2m} dp_y \int_{-\infty}^{\infty} e^{-\beta p_z^2/2m} dp_z$$

$$Z = \frac{V}{h^3} \left(\frac{2\pi\kappa}{\beta} \right)^{3/2}$$

$$= \frac{V}{h^3} (2\pi m k_B T)^{3/2}$$

$$= V \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2}$$

$$Z = \frac{V}{h^3}$$

$$\text{where } l = \frac{h}{\sqrt{2\pi m k_B T}}$$

\downarrow
De-Broglie wavelength.

Q. Write the partition funⁿ for a particle of mass m whose pot. energy is given.

$$V(x, y, z) = ax^2 + b(y^2 + z^2)^{1/2}$$

Solⁿ:

$$E = \frac{p^2}{2m} + V$$

$$= \frac{1}{2m}(p_x^2 + p_y^2 + p_z^2) + ax^2 + b(y^2 + z^2)^{1/2}$$

$$Z = \frac{1}{h^3} \iiint e^{-\beta \left[\frac{1}{2m}(p_x^2 + p_y^2 + p_z^2) + ax^2 + b(y^2 + z^2)^{1/2} \right]} dx dy dz dp_x dp_y dp_z$$

$$= \frac{1}{h^3} \int_{-\infty}^{\infty} e^{-\beta ax^2} dx \int_{-\infty}^{\infty} e^{-\beta p_x^2/2m} dp_x$$

$$= \frac{1}{h^3} \sqrt{\frac{\pi}{\beta a}} \left(\frac{2m\pi}{\beta} \right)^{3/2} 2\pi \left(\frac{1}{\beta b} \right)^2$$



Q. In a system of particle fixed in space, each particle has total q. No. J, the mag. q. no. m_j , which has $(2J+1)$ values $-J, (-J+1), \dots, +J$.

In presence of mag. field B, the energy of a level is given by $E = -m_j \mu_B B q$.

$q \rightarrow$ Landé g-factor

find partition funⁿ for such an atom in mag. field

Solⁿ:

$$Z = \sum g_i e^{-\beta E_i}$$

$$= \sum_{m_j=-J}^{+J} 1 \cdot e^{+\beta m_j \mu_B B q}$$

$$= e^{-\beta J \mu_B B q} + e^{\beta(-J+1) \mu_B B q} + \dots + e^{\beta(J+1) \mu_B B q}$$

G.P.
 $\frac{a(1-r^n)}{1-r}$

$$= \frac{\sinh(\beta \mu_B B q (J+1))}{\sinh(\frac{\beta \mu_B B q}{2})}$$

$$\begin{aligned}
 Z &= e^{-\beta J M_B B q} [1 + e^{-\beta B q} + e^{-\beta B q} + \dots + e^{2 \beta M_B B q}] \\
 &= \frac{e^{-\beta J M_B B q} [e^{\beta M_B B q (2J+1)} - 1]}{e^{\beta M_B B q} - 1} \\
 &= \frac{e^{\beta M_B B q (J+1)} - e^{-\beta J M_B B q}}{e^{\beta M_B B q / 2} [e^{\beta M_B B q / 2} - e^{-\beta M_B B q / 2}]} - \frac{\sinh(\beta M_B B q (J+1))}{\sinh(\beta M_B B q / 2)}
 \end{aligned}$$



* for a system of N distinguishable particle the partition funⁿ.

$$Z_{\text{system}} = (z_i)^N$$

Prob: Consider a 1-D Ising model of N spins at very low temp. where all spins are aligned $\parallel e$ to each other. There will be few spins flip at temp. T . With each spin flip $2J$ energy is added.

In a configuration of N spins, r spins are flip, the energy of the system is $\sum_{\text{all spin are } \parallel} E = -N J + 2r J$ (flip spin energy)

No. of configuration is $N_C r$, r varies from 0 to N . The partition funⁿ of the system is.

$$\text{Soln: } Z = \sum_r g_r e^{-\beta E_r}$$

$$\begin{aligned}
 &= \sum_{r=0}^N N_C r e^{-\beta (-N J + 2r J)} \\
 &= (2 \cosh \beta J)
 \end{aligned}$$

Boltzmann's definition of probability:

$$P(E_i) = \frac{n_i}{N} = \frac{g_i e^{-\beta E_i}}{\sum_i g_i e^{-\beta E_i}}$$

$$N_1 = \frac{N_0 e^{-\beta E_1}}{Z}$$

$$N_1 = \frac{N_0}{e^{\beta E_1}}$$

Prob: In a system, there are three energy states, having energies, $0, K_B T$ & $2K_B T$. The total energy of the system is $1000 K_B T$. find the no. of particle in different state & average en. of each particle.

Soln: N_2 ————— $2K_B T$ find N_0, N_1, N_2

N_1 ————— $K_B T$

N_0 ————— 0



$$\begin{aligned} H &=? \\ E &=? \end{aligned}$$

deco formulas

$$N = N_0 + N_1 + N_2, \quad N = N_0 + \frac{N_0}{e^1} + \frac{N_0}{e^2}$$

~~$$\frac{N_1 (E = K_B T)}{N_0} = \frac{N e^{-B K_B T / Z}}{N e^{-0 / Z}}$$~~

$$N_0 = \frac{N}{Z}$$

$$N_1 = \frac{N e^{-1}}{Z}$$

$$\frac{N_2}{N_1} = \frac{g_2 e^{-(E_2 - E_1) \beta}}{g_1}$$

$$E = 1000 K_B T = N_0 E_0 + N_1 E_1 + N_2 E_2$$

$$= N_0 \times 0 + N_1 K_B T + N_2 2 K_B T$$

$$1000 = N_1 + 2N_2$$

$$= \frac{N_0}{e} + \frac{2N_0}{e^2} \Rightarrow N_0 = \frac{1000 e^2}{(e^2 + 2)}$$

$$\overline{E} = \frac{E}{N} = \frac{K_B T (2 + e)}{(1 + e + e^2)}$$

Prob: A system of 5 identical but distinguishable particles having energy 3ϵ . The single particle state are available at energies, $0, \epsilon, 2\epsilon, 3\epsilon$.

- Find the ave no. of particles in each energy state
- Sketch $n(\epsilon)$ with energy.

Soln,

En. of the system	Energy of the state				No. of ways of distribution
	0	ϵ	2ϵ	3ϵ	
3ϵ	4	0	0	1	$\frac{15}{4!0!0!1!} = 5$
	3	1	1	0	20
	2	3	0	0	10

i)

$$n(0) = n_1 p_1 + n_2 p_2 + \dots$$

$$= 4 \times \frac{5}{35} + 3 \times \frac{20}{35} + 2 \times \frac{10}{35}$$

$$= 2.8$$

$$20 + 10 + 5 = 35$$

Probabilities

$$\frac{5}{35}$$

$$\frac{20}{35}$$

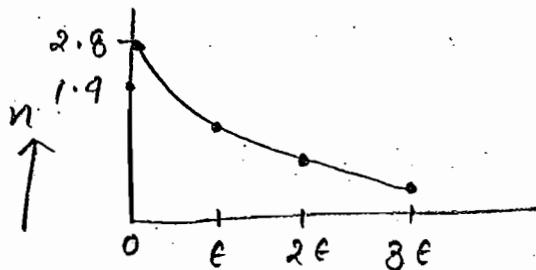
$$\frac{10}{35}$$

$$n(\epsilon) = 1.4$$

$$n(2\epsilon) = 0.571$$

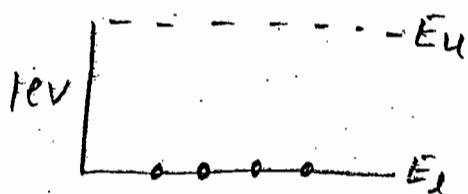
$$n(3\epsilon) = 0.14$$

ii)



Ques: The energy required to create a lattice vacancy in a crystal is 1 eV. find out the ratio of vacancies at temp 1200K & 300K. a.) e^{-30} b.) e^{30} c.) e^{-10} d.) e^{10}

Soln:



$$n(T) \propto e^{-\Delta E/k_B T}$$

$$n(1200) \propto e^{-1eV/k_B \times 1200}$$

$$n(300) \propto e^{-1eV/k_B \times 300}$$

ask

$$\frac{n(1200)}{n(300)} = \frac{e^{-1eV/k_B \times 1200}}{e^{-1eV/k_B \times 300}} = \frac{e^{-1200}}{e^{-300}}$$



Q. The states with energy diff'n $4.83 \times 10^{-21} J$ occurs w/ relative probability e^2 . find temp. (Ans 175K)

Ques: A system can take only 3 different energy states of energy $E_1 = 0$, $E_2 = 1.38 \times 10^{-21} J$, $E_3 = 2.76 \times 10^{-21} J$.

These 3 states occur in 2, 5, 4 different ways respectively. find the probability that at 100K, the system may be

1.) In one of 4-state of energy E_3 $P(E_3) = ?$ $\frac{4e^{-\beta E}}{Z}$

2.) In the or.s. $P(E_1) = \frac{2e^{-\beta E}}{Z}$

$$\textcircled{1} \quad \frac{P(E_1)}{P(E_2)} = e^2 = e^{\Delta E/k_B T}$$



Q. A system has two masses with vibrations with frequency of vibrations with ω , & $\omega_2 (=2\omega_1)$. Let V is the pot. en. of the system. Also find the probability that at temp. T , the system has energy less than $4\hbar\omega_1$.

Soln: $V = \alpha x^2 + 4\alpha y^2$

OR

$$V = \frac{1}{2} m \omega_1^2 (x^2 + 4y^2)$$

$$E_{nx, ny} = \left(n_x + \frac{1}{2}\right)\hbar\omega_1 + \left(n_y + \frac{1}{2}\right)\hbar\omega_2 \rightarrow 2\hbar\omega_1$$

$$= \left(n_x + 2n_y + \frac{3}{2}\right)\hbar\omega_1$$



$$P(< 4\hbar\omega_1) = \frac{1 \cdot e^{-\beta \frac{3}{2}\hbar\omega_1} + 1 \cdot e^{-\beta \frac{5}{2}\hbar\omega_1} + 2e^{-\beta \frac{7}{2}\hbar\omega_1}}{Z}$$

$$n_x \quad n_y = \frac{e^{-\frac{3}{2}\beta\hbar\omega_1}}{Z} \left(1 + e^{-\beta\hbar\omega_1} + e^{-2\beta\hbar\omega_1}\right)$$

$$\frac{3}{2}\hbar\omega_1 \quad 0 \quad 0 = \frac{x^{3/2} (1+x+2x^2)}{Z}$$

$$\frac{5}{2}\hbar\omega_1 \quad 1 \quad 0 = \frac{1}{Z}$$

$$\frac{7}{2}\hbar\omega_1 \quad 2 \quad 0 \quad \left\{ \text{degeneracy} \right.$$

$$x = e^{-\beta\hbar\omega_1}$$

Ensembles: Gibbs

→ A collection of large no. of macroscopically identical but essentially independent system, is called ensemble.

OR

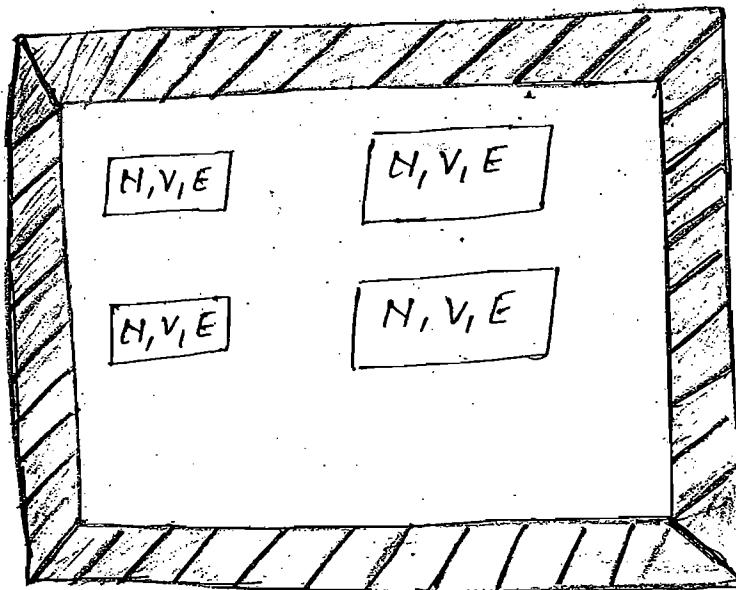
→ A system is defined as "coll. of large no. of particles"

Macroscopic identical : System having the same macroscopic quantities like, T, V, N, E, μ, P etc.

→ If we have a coll. of large no. of independent system having macroscopic properties 1.) N, V, E same \rightarrow microcanonical ensemble
 more restriction
 2.) T, V, N same \rightarrow canonical ensemble
 less restriction
 3.) T, V, μ same
 \rightarrow Grand canonical ensemble
 No restriction.

System → Involving in time \rightarrow We take the time averaged quantity.

$$\langle \text{Time ave behaviour} \rangle = \langle \text{ensemble averaged behaviour} \rangle$$



MCE

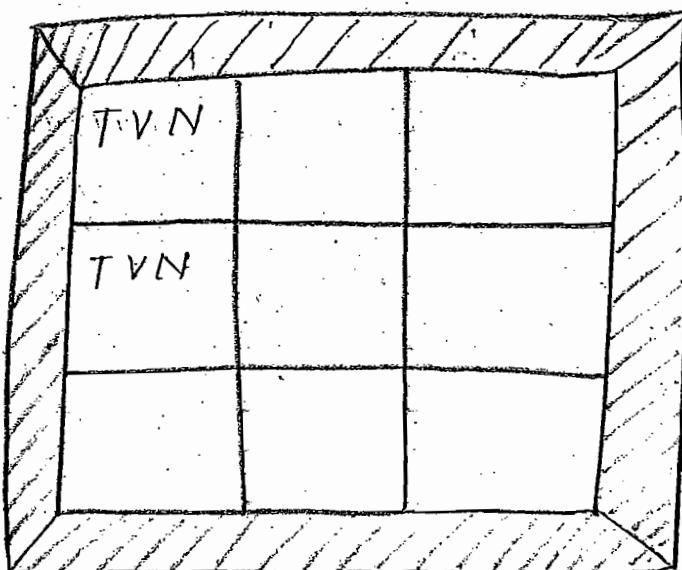
* Ensemble average of any property would be very close to time average value of that property over a single system.

Properties

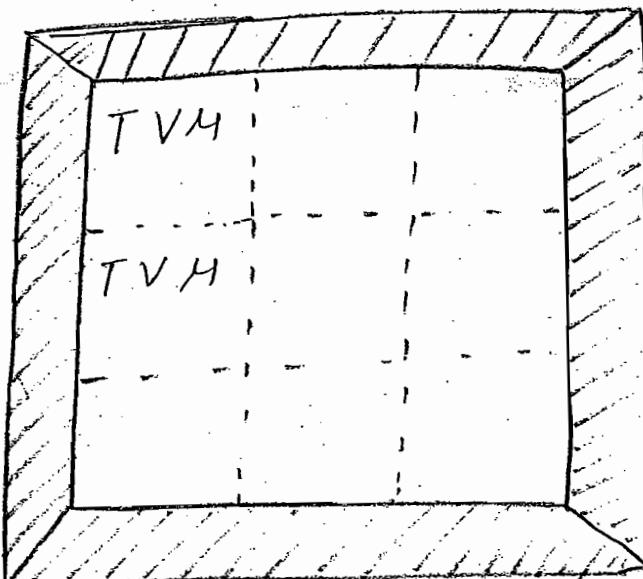
Contact with the environment

MCE

No contact i.e isolated.
i.e., No exchange of Energy & matter
(particle). i.e., walls should be
rigid, impenetrable & not conducting
($V=0$) $dN = 0$



Canonical Ensembles



G.O.E.

Properties	M.C.E.	C.E.	G.C.E.
Contact with the environment	No contact ie isolated.	System in thermal contact with heat reservoir. Walls should be conducting, rigid & impenetrable.	System can exchange both energy & matter.
fluctuations	None	fluctuation in energy only	fluctuation in energy & no. of particle both.
Density function $P(E) = \frac{N}{V} = \text{constt}$ $P(E)$	$P(E) = Ae^{-E/k_B T}$		$P(E) = e^{(N_f + N_A - E)/k_B T}$
Partition function (Z)	$Z = \Delta T$	$Z = \int e^{-\beta E} d\tau$	$Z = \sum_n e^{UN/T} \cdot Z_n$
What is the use of Ensembles → Internal energy of an ideal mono atomic gas $\rightarrow \frac{3}{2} NKT$			

Average properties of different quantities →

\bar{E} , $\bar{E^2}$, \bar{P} , \bar{H} , ΔE etc

$$\rightarrow \bar{X} = \frac{n_1 x_1 + n_2 x_2 + \dots}{n_1 + n_2 + \dots}$$

$$= \frac{n_1}{N} x_1 + \frac{n_2}{N} x_2 + \dots$$

$$= P_1 x_1 + P_2 x_2 + \dots$$

$$\bar{X} = \sum_i P_i x_i$$

$$\rightarrow \bar{E} = \sum_i P_i \epsilon_i$$

$$= \sum_i \frac{e^{-\beta \epsilon_i}}{Z} \epsilon_i$$

$$= \sum_i \frac{e^{-\beta \epsilon_i}}{\sum e^{-\beta \epsilon_i}} \epsilon_i$$

$$\boxed{\bar{E} = -\frac{\partial}{\partial \beta} \ln Z = \frac{1}{Z} \frac{\partial Z}{\partial \beta}}$$

OR

$$\bar{E} = K_B T^2 \frac{\partial \ln Z}{\partial T} \quad \text{where } \beta = \frac{1}{K_B T}$$

$$\rightarrow \bar{E^2} = \sum_i P_i \epsilon_i^2$$

$$= \sum_i \frac{e^{-\beta \epsilon_i}}{\sum e^{-\beta \epsilon_i}} \epsilon_i^2$$

$$\boxed{\bar{E^2} = \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2}}$$



$$\rightarrow (\overline{\Delta E})^2 = (\overline{E^2}) - (\overline{E})^2$$

$$= \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2} - \left(\frac{1}{Z} \frac{\partial Z}{\partial \beta} \right)^2$$

$$(\overline{\Delta E})^2 = \frac{\partial^2}{\partial \beta^2} \ln Z$$

OR

$$(\overline{\Delta E})^2 = - \frac{\partial}{\partial \beta} \left(- \frac{\partial}{\partial \beta} \ln Z \right)$$

$$= - \frac{\partial}{\partial \beta} (\overline{E})$$

$$= K_B T^2 \frac{\partial \overline{E}}{\partial T}$$

$$= K_B T^2 \left(\frac{3}{2} N K_B \right)$$

for ideal gas

$$(\overline{\Delta E})^2 = K_B T^2 C_V$$

$$\boxed{\Delta E = K_B T \sqrt{\frac{3}{2} N}}$$

$$\frac{\overline{\Delta E}}{\overline{E}} = \frac{K_B T \left(\frac{3}{2} N \right)^{1/2}}{\frac{3}{2} N K_B T}$$

$$\frac{\overline{\Delta E}}{\overline{E}} = \left(\frac{3}{2} N \right)^{-1/2} = \left(\frac{2}{3N} \right)^{1/2}$$



Relative fluctuation in energy

$$\frac{\Delta E}{E} \propto N^{-1/2} \text{ or } \frac{1}{\sqrt{N}} \quad 10^{22} \text{ atoms}$$

$$\propto 10^{-11}$$

$\rightarrow 0$, In ideal gas, fluctuations in en. is approx to zero.

Average Pressure:

$$\bar{p} = \sum_i p_i P_i$$

$P_i \rightarrow$ Probability

$p_i \rightarrow$ Pressure

$$\bar{p}_i = \sum_i \frac{e^{-\beta E_i}}{\sum_i e^{-\beta E_i}} \left(-\frac{\partial E_i}{\partial V} \right)_N \quad -①$$

We KIT

$$TdS = dE + pdV - HdN$$

At equilibrium
 $TdS \rightarrow 0$

or Entropy fixed state

$$p = \left(-\frac{\partial E}{\partial V} \right)_N$$

from eqⁿ ①

$$\bar{p} = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} \Big|_{T, N} \quad \text{Baronett}$$

OR

$$= \frac{1}{\beta} \frac{1}{Z} \frac{\partial Z}{\partial V} \Big|_{T, N}$$

$$\bar{p} = \frac{N}{\beta V} = \frac{Nk_B T}{V}$$

$$\bar{p} = \frac{2}{3} \frac{E}{V}$$



Average chemical potential \rightarrow

$$\begin{aligned}\bar{\mu} &= \sum_i p_i \mu_i \\ &= \frac{\sum_i e^{-\beta E_i} \left(\frac{\partial E_i}{\partial N} \right)_V}{Z}\end{aligned}$$

$$\boxed{\bar{\mu} = -\frac{1}{\beta} \frac{\partial}{\partial N} \ln Z / V, T}$$

Helmholtz free energy (F) and Entropy in terms of Z \rightarrow

$$F = U - TS$$

$$dF = dU - Tds - SdT$$

use 1st law: $Tds = dU + Pdv$

Then

$$dF = -Pdv - SdT$$

$$\rightarrow \text{If } dv = 0$$

$$dT = 0$$

$$dF = 0$$

when $F \rightarrow \min^m$ energy, then system is in equilibrium.

$$d[\ln Z(T, V, N)] = \frac{\partial \ln Z}{\partial \beta}_{V, N} d\beta + \frac{\partial \ln Z}{\partial V}_{T, N} dV + \frac{\partial \ln Z}{\partial N}_{T, V} dN$$

$$d[\ln Z(T, V, N)] = -E d\beta + \beta P dV - \beta \bar{\mu} dN \quad \text{--- (1)}$$

$$d(\beta E) = \beta dE + E d\beta \quad \text{--- (2)}$$

Add (1) + (2)

$$d[\ln Z + \beta E] = \beta [dE + PdV - \lambda dN]$$

$$= \beta T ds$$

$$= \frac{1}{k_B} ds$$

$$ds = d[K_B \ln Z + k_B \beta E]$$

$$S = K_B \ln Z + \frac{E}{T}$$

gmp

$$F = E - TS = -K_B T \ln Z$$



$$\rightarrow F = E - TS$$

$$dF = dE - Tds - sdt$$

$$= dE - (dE + PdV - \lambda dN) - sdt$$

$$df = -PdV + \lambda dN - sdt$$

From this relⁿ, we can find that

$$P = \left. -\frac{\partial F}{\partial V} \right|_{N, T}$$

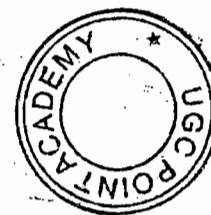
$$M = \left. \frac{\partial f}{\partial N} \right|_{V, T}$$

$$\text{Entropy } S = \left. -\frac{\partial f}{\partial T} \right|_{N, V}$$

$$\rightarrow E = F + TS$$

$$E = F - T \left. \frac{\partial F}{\partial T} \right|_{N, V}$$

Helmholtz free energy.



Prob.

1- DCH₃O

$$E = \frac{p^2}{2m} + \frac{1}{2} Kx^2$$

$$\text{The partition funn } Z = \frac{K_B T}{\hbar \omega}$$

find ave thermal En.

Soln:

$$\langle E \rangle = -\frac{\partial \ln Z}{\partial \beta}$$

$$= -\frac{\partial}{\partial \beta} \ln \left(\frac{1}{\beta \hbar \omega} \right)$$

$$= -\beta \hbar \omega \times \frac{1}{\hbar \omega} \times -\frac{1}{\beta^2}$$

$$= \frac{1}{\beta}$$

$$= K_B T$$

$$\langle E \rangle = \frac{1}{2} K_B T + \frac{1}{2} K_B T$$

- Q. The hamiltonian of a particle of classical ideal gas is $H = \frac{1}{2m}(px^2 + py^2 + pz^2)$

alt is the ave. energy of the molecule.

Ans $\frac{3}{2} K_B T$

Soln: We KIT

$$Z = \frac{V}{h^3} = \frac{V}{h^3} (2\pi m K_B T)^{3/2}$$

- Q. A system can have three energy levels $E=0, \pm \epsilon$. Level $E=0$ is doubly degenerate while others are non-degenerate. Find the ave. energy of the system.
- a) $-\epsilon \tanh \beta \epsilon / 2$ b) $-\epsilon \tanh \beta E$ c) $\epsilon \cosh \beta E$

Soln:

$$\text{Partition function } Z = 2e^{-0} + e^{-\beta E} + e^{\beta E}$$

$$= 2 + e^{\beta E} + e^{-\beta E}$$

$$= (e^{\beta E/2} + e^{-\beta E/2})^2$$

$$= (2 \cosh \beta E/2)^2$$

$$\bar{E} = -\frac{\partial \ln Z}{\partial \beta}$$



$$= -\frac{\partial}{\partial \beta} \ln (2 \cosh \beta E/2)^2$$

$$= -\frac{2}{Z} \cdot \frac{1}{\cosh \frac{\beta E}{2}} \cdot \sinh \beta E/2 \cdot \beta E/2$$

$$= -\tanh \frac{\beta E}{2} \cdot \epsilon$$

$$\bar{E} = -\epsilon \tanh \beta \epsilon / 2 \quad \text{Ans}$$

- Q. Av. thermal En: 1-D q. osc.

$$Z = \frac{e^{-\beta \hbar \omega / 2}}{1 - e^{-\beta \hbar \omega}}$$

$$\bar{E} = -\frac{\partial}{\partial \beta} \ln Z / v, N$$

$$= \left(\frac{1}{2} \hbar \omega \right)^2 + \frac{\hbar \omega}{e^{\beta \hbar \omega} - 1}$$

$$\bar{E} = \frac{\hbar \omega}{e^{\hbar \omega / k_B T} - 1}$$

* $P = E/C, E = h\nu$
 $\omega = 2\pi\nu = 2\pi \frac{C}{T} = K_C$

$$\omega = \frac{E}{\hbar}, K = \hbar / k$$

$$\frac{\hbar}{2\pi} \frac{2\pi\nu}{e^{\frac{\hbar \nu}{k_B T} - 1}}$$

Q. If gas or impurity occupies in a volume V at temp. T .
The logarithm of partition funn' is given by

$$\ln Z = N \ln [(V - bN)(K_B T)^{3/2}]$$

find (i) Avg internal En $\frac{3}{2} N K_B T$

(ii) Eqn of state of the gas.

Soln: (ii) $p = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} |_{N, T}$

$$p = \frac{N K_B T}{V - bN}$$

$$p(V - bN) = N K_B T \quad (\text{semi-vander-wal eqn})$$

(i) $\bar{E} = - \frac{\partial \ln Z}{\partial \beta}$

$$= - \frac{\partial}{\partial \beta} [N \ln (V - bN)(K_B T)^{3/2}]$$

Q. The partition funⁿ for a system of N particles is

$$VN \left(\frac{2\pi m k_B T}{h^2} \right)^{3N/2}$$

find

$$(i) \langle E \rangle = \frac{3}{2} N k_B T$$

$$(ii) \text{Eq}^n \text{ of state } PV = N k_B T$$

Q. A gas of N non-interacting particle is in thermal equilibrium at temp. T . Each particle can be any of the possible non-degenerate states of energy $0, 2E$ & $4E$. The average energy per particle of the gas, when βE is $2E, 3E, 2E/3, E$.

Q. The partition funⁿ of 1-D H-O. is

$$Z = \frac{e^{-\beta \hbar \omega / 2}}{1 - e^{-\beta \hbar \omega}} \quad \langle n \rangle = \frac{1}{e^{\beta \hbar \omega} - 1}$$

find the average no. of oscillations or quanta.

$$\text{Ans} \quad Z = 1 + e^{-2\beta E} + e^{-4\beta E}$$

$$\langle E \rangle = \frac{-\partial}{\partial \beta} \ln Z = \frac{0 + 2E e^{-2\beta E} + 4E e^{-4\beta E}}{1 + e^{-2\beta E} + e^{-4\beta E}}$$

$$\beta E \ll 1$$

$$\frac{4E}{3} = 2E$$

Ave En. of System.

$$(3) \text{ Ave Osc. } \langle n \rangle = \sum_{n=0}^{\infty} \frac{n e^{-\beta E_n}}{Z} \quad E_n = \left(n + \frac{1}{2}\right) \hbar \omega$$

$$n = 0, 1, 2$$

$$= 0 e^{-\beta \hbar \omega / 2} + 1 \cdot e^{-\frac{3}{2} \hbar \omega} + 2 e^{-\frac{5}{2} \hbar \omega} + \dots$$

$$\langle n \rangle = \frac{e^{-\frac{3}{2} \hbar \omega} (1 + 2e^{-\beta \hbar \omega} + 3e^{-2\beta \hbar \omega} + \dots)}{Z}$$



$$1+2x+3x^2 + \dots = \frac{(1-x)^{-2}}{1-x} = (1-x)^{-1}$$

$$\langle n \rangle = \frac{e^{-\frac{3\hbar\omega}{2}}}{z} \frac{1}{(1-e^{-\beta\hbar\omega})^2}$$

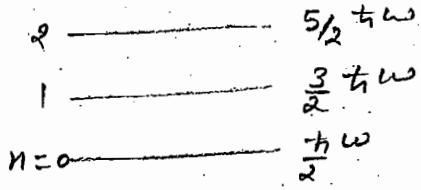
$$= \frac{e^{-\beta\hbar\omega}}{1-e^{-\beta\hbar\omega}}$$

$$\langle n \rangle = \frac{1}{e^{\beta\hbar\omega}-1}$$

$$\langle n \rangle = \frac{1}{e^{\hbar\omega/k_B T}-1}$$

Ave. No. of photon or phonons
or quanta.

→ Oscillator emits & absorbs energy in the form of
quanta.



$$E_n = (n+\frac{1}{2})\hbar\omega$$

$$E_n = n\hbar\nu$$

$$E_n = 0, \hbar\nu, 2\hbar\nu \dots$$

$$\text{Ave. No. of qu.} \left[\begin{array}{l} \\ \\ \\ \end{array} \right] = 0, 1, 2, 3 \dots$$

Total No. of Oscillator:

$$N = N_0 + N_1 + N_2 + \dots$$

2 N_2
 1 N_1 $\frac{3\hbar\omega}{2}$
 n=0 N_0 $\frac{1\hbar\omega}{2}$

$$= N_0 + N_0 e^{-\beta\hbar\omega} + N_0 e^{-2\beta\hbar\omega} + \dots$$

$$N = \frac{N_0}{1 - e^{-\beta\hbar\omega}}$$

Total No. of Osc.

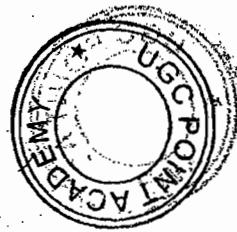
$$\begin{aligned}
 E &= N_0 E_0 + N_1 E_1 + \dots \\
 &= N_0 \left(\frac{1}{2}\hbar\omega\right) + N_0 e^{-\beta\hbar\omega} \left(\frac{3}{2}\hbar\omega\right) + \dots \\
 &= \frac{1}{2} N_0 \hbar\omega \left[1 + 3e^{-\beta\hbar\omega} + 5e^{-2\beta\hbar\omega} + \dots \right] \\
 &= \frac{1}{2} \frac{N_0 \hbar\omega}{(1 - e^{-\beta\hbar\omega})^2}
 \end{aligned}$$

$$\text{Av. En} = \frac{N \hbar\omega}{e^{\beta\hbar\omega} - 1} = N \langle E \rangle =$$

→ If G is given

$$G = N k_B T (1 - \ln z)$$

$$G = E + PV - TS$$



Q. $Z = \frac{e^{-\beta \hbar \omega / 2}}{1 - e^{-\beta \hbar \omega}}$

Calculate Av. no. of oscillators.

Solⁿ: Av. En. of osc = $\frac{\hbar \omega}{e^{\hbar \omega / k_B T} - 1}$

Av. no of osc = $\frac{1}{e^{\hbar \beta \omega} - 1}$



Q. for a non-relativistic fermi gas

$$E \propto \frac{N^{5/3}}{V^{2/3}}$$

The pressure of gas is.

$$\frac{1}{3} \frac{E}{V}, \frac{2}{3} \frac{E}{V}, \frac{5}{2} \frac{E}{V}$$

Solⁿ:

$$\overline{P} = \left(-\frac{\partial E}{\partial V} \right)_N$$

$$\overline{P} = \frac{2}{3} \frac{E}{V}$$

Q. for a certain system the energy of each state is given by

$$E_S = f_S(T) - N k_B T \ln \left(\frac{V}{V_0} \right)$$

$$V_0 = \text{constt}$$

(i) Write the partition funⁿ of the system.

(ii) Av. Pressure of the system

(iii) \overline{H}

$$Z = \sum_S e^{-\beta E_S} \xrightarrow{\text{value}}$$

$$\overline{P} = \frac{N k_B T}{V}, \quad \overline{H} = -\frac{1}{\beta} \frac{\partial \ln Z}{\partial N} |_{T, V}$$

$$\underline{\text{Sol'n:}} \quad Z = \sum_s e^{-\beta E_s} = \left(\frac{V}{V_0}\right)^N \sum_s e^{-\beta E_s(N)}$$

$$= \sum_s e^{-\beta N E_s}$$

$$H = -\frac{1}{\beta} \frac{\partial}{\partial N} \left[N \ln \left(\frac{V}{V_0} \right) \right]$$

$$= -\frac{1}{\beta} \ln \left(\frac{V}{V_0} \right)$$

$$= -K_B T \ln \left(\frac{V}{V_0} \right)$$

$$\bar{P} = \frac{N K_B T}{V}$$

Q. for a certain system

$$Z(T, V, N) = \left(\frac{P}{P_0}\right)^{\frac{3N}{2}} \left(\frac{V}{V_0}\right)^N$$

find \bar{E} , F , \bar{P} , \bar{H} , \bar{s} , eqⁿ of state, & $\Delta E / \bar{E}$

Sol'n:





Q. The partition funⁿ for a certain system is given as

$$Z = \left(\frac{B}{B_0}\right)^{-N} e^{-N\left(\frac{V}{V_0}\right)} - 1^0 , B_0, V_0 \text{ are const}$$

find $\bar{E}, F, P, \bar{H}, S$, eqⁿ of states. AEIE



Average thermal energy of 1-v classical gas \rightarrow

Some important formula:

$$\cancel{H = E + PV}$$

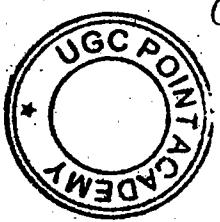
$$= NK_B T^2 \left[\frac{\partial}{\partial T} (\ln z) \right]_V + NK_B T$$

$$G_f = H - TS$$

$$= NK_B T^2 \left[\frac{\partial}{\partial T} (\ln z) \right]_V + NK_B T - NK_B \ln z - E$$

$$G_f = K_B T (1 - \ln z)$$

Specific heat



$$C_V = T \frac{\partial S}{\partial T} \Big|_V$$

$$= -T \left(\frac{\partial^2 f}{\partial T^2} \right)_V$$

Isothermal compressibility

$$K_T = \frac{1}{B} = -V \left(\frac{\partial P}{\partial V} \right)_T$$

$$= V \left[\frac{\partial^2 f}{\partial V^2} \right]_T$$

$$= \left[V \frac{\partial^2}{\partial V^2} (-NK_B T \ln z) \right]_T$$

Q. The partition function for an interacting gas (real) is assumed to

$$Z = \left(\frac{V-Nb}{N} \right)^N \left(\frac{mKBT}{2\pi \hbar^2} \right)^{3N/2} e^{N^2 a^2 / V KBT}$$

a, b are constt

Show that the pressure of the gas

$$P = \frac{NKBT}{(V-Nb)} - \frac{N^2 a^2}{V^2} \Rightarrow \left(P + \frac{N^2 a^2}{V^2} \right) (V-Nb) = NkT$$

Soln:



Q. A system consisting of non-interacting classical particles of spin $\frac{1}{2}$ & mag. moment M , each kept in constt. external m.f. is in normal equilibrium at temp. T . The magnetisation of the system is $NM \coth\left(\frac{MH}{k_B T}\right)$, $NM \tanh\left(\frac{MH}{k_B T}\right)$.

Solⁿ:

$$M = -\frac{\partial F}{\partial H} \Big|_{T, V, N}$$

$$F = -Nk_B T \ln Z$$

$$= -Nk_B T \ln \left[2 \cosh\left(\frac{MH}{k_B T}\right) \right]$$

$$M = NM \tanh\left(\frac{MH}{k_B T}\right)$$

$$Z = \sum e^{-\beta E_i}$$

$$= e^{-\beta(MH)} + e^{\beta MH}$$

$$Z = 2 \cosh \beta M H$$

Q. An ensemble of N , three level system with energies $-E_0, 0, +E_0$ in thermal equilibrium at temp. T . If $\beta E_0 = 2$. What is the prob. of finding the system in the level $E=0$? Probability $P = \frac{g_0 e^{-\beta E_0}}{\sum g_i e^{-\beta E_i}}$

$$\frac{\cosh 2}{2}, (\cosh 2)^{-1}, (2 \cosh 2)^{-1}, (1 + 2 \cosh 2)^{-1}$$

Q. $E = \frac{bs^3}{VN}$, $T = ?$ $\frac{bs^2}{VN}$, $\frac{3bs^2}{VN}$ etc.

Solⁿ:

$$Z = \sum_i e^{-\beta E_i} = 1 + e^{-\beta E_0} + e^{\beta E_0}$$

$$P = \frac{g_0 e^{-\beta E_0}}{\sum_i g_i e^{-\beta E_i}} \Rightarrow \frac{e^{-\beta E_0}}{\sum_i e^{-\beta E_i}} = \frac{1}{1 + e^{-2} + e^2}$$

$$= \frac{1}{1 + \left(\frac{e^{-2} + e^2}{2}\right)^{1/2}}$$

$$= \frac{1}{1 + 2 \cosh 2} \\ = (1 + 2 \cosh 2)^{-1}$$

griff

$$T = \frac{\partial E}{\partial S} \Big|_{V, N}$$

$$dT = TdS$$

$$\checkmark TdS = dE + PdV - \mu dN$$

$$\checkmark dF = -SdT - PdV + \mu dN$$



Average thermal energy of 1-D classical O.G. →

1-D Cl. H.O.

$$E = \frac{Px^2}{2m} + \frac{1}{2} Kx^2$$

$$\langle E \rangle = \frac{1}{2m} \langle Px^2 \rangle + \frac{1}{2} K \langle x^2 \rangle$$

$$= \frac{1}{2m} m K_B T + \frac{1}{2} K_B T$$

$$= \frac{1}{2} K_B T + \frac{1}{2} K_B T$$

$$\langle E \rangle = K_B T$$

→ Average thermal energy corresponding to each quadratic term of momentum & position is $\frac{1}{2} K_B T$. (Equipartition theorem)

$$\langle E \rangle = \frac{\int H(q, p) e^{-\beta H(q, p)} dq dp}{\int e^{-\beta H(q, p)} dq dp}$$

If

$$\bar{x} = \sum x_i p_i$$

$$= \frac{\sum x_i e^{-\beta H(q, p)}}{\sum e^{-\beta H(q, p)}}$$

$$\langle E \rangle = \frac{\iiint \left(\frac{Px^2}{2m} + \frac{1}{2} Kx^2 \right) e^{-\beta \left(\frac{Px^2}{2m} + \frac{1}{2} Kx^2 \right)} dx dp}{\iint e^{-\beta \left(\frac{Px^2}{2m} + \frac{1}{2} Kx^2 \right)} dx dp}$$

unit

$dq dp \rightarrow \text{Joule} \cdot \text{sec}$

Average thermal energy $\text{v} = \nu \cdot u = 0.005 \rightarrow$

$$\langle E \rangle = 3k_B T$$

Q. A system has N particle having p.e. $\alpha x^2 + \beta y^2$. The Av. thermal En. of the system is

$$= 2Nk_B T$$

$$H = \frac{1}{2m}(p_x^2 + p_y^2) + \alpha x^2 + \beta y^2$$

$$\langle E \rangle = 2k_B T$$

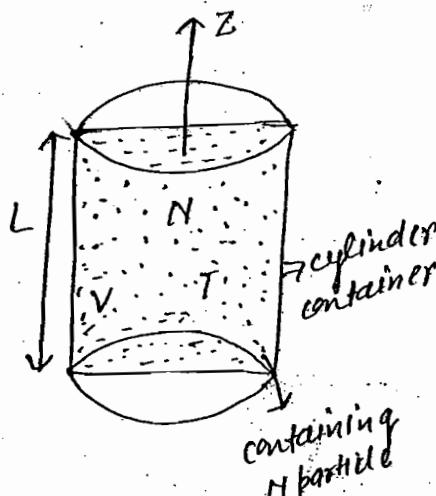


* Non-interacting cl. ideal gas having N particles, Av. thermal En.

$$\langle E \rangle = \frac{3}{2} Nk_B T$$

Q. A system of N cl. non-interacting particles each of mass m at temp. T and is confined by the external pot. $V = \frac{1}{2} \pi R^2$, where $R = \text{constt.}$ in 3-D find the Av. Int. En. of the system.

$$\frac{3}{2} Nk_B T, \frac{5}{2} Nk_B T, 3Nk_B T, \frac{1}{2} Nk_B T$$



$L \ll R$

$$mgL \gg k_B T$$

$\langle E \rangle$ of the system

$$\frac{3}{2} Nk_B T, \frac{5}{2} Nk_B T$$

Sol'n: Hamiltonian of the particle

$$H = \frac{1}{2m}(p_x^2 + p_y^2 + p_z^2) + mgz$$

$$\frac{3}{2} k_B T$$

$\langle E \rangle = \iiint \text{He}^{-\beta H} dx dy dz dp_x dp_y dp_z$

$$\iiint e^{-\beta H} dx dy dz dp_x dp_y dp_z$$



Q. A particle confined in a region $x_{1,0}$ by a pot. whose value increases with x as $V(x) = \frac{1}{2}kx^2$, k_0 constt. Find mean position of the particle at temp T.

$$\frac{K_B T}{4_0}, \frac{K_B T}{2H_0} \text{ etc.} \quad \int_0^\infty t^4 e^{-at} dt = \frac{4!}{a^5}$$

Soln:

$$\langle x \rangle = \frac{\iint x e^{-\beta H(q,p)} dq dp}{\iint e^{-\beta H(q,p)} dq dp}$$

$$= \frac{\int_0^\infty x e^{-\beta \frac{1}{2}kx^2} dx}{\int_0^\infty e^{-\beta \frac{1}{2}kx^2} dx}$$



Thermal Expansion of solids:

Avg. value of thermal expansion

$$\langle x \rangle = \frac{\iint x e^{-\beta H} dx dP_x}{\iint e^{-\beta H} dx dP_x}$$

cl. H.O. treatment

$$H = \frac{p x^2}{2m} + \alpha x^2 \quad \text{H.O.}$$

Parabolic pot.

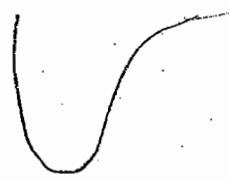
$$\langle x \rangle = \frac{\int_{-\infty}^{\infty} x e^{-\beta ax^2} dx}{\int_{-\infty}^{\infty} e^{-\beta ax^2} dx} = 0$$



cl. An H.O. treatment:

$$V(x) = \alpha x^2 - b x^3 - c x^4$$

Anharmonic terms
 b, c are smaller than α



$$\langle x \rangle = \frac{\int_{-\infty}^{\infty} x e^{-\beta(\alpha x^2 - b x^3 - c x^4)} dx}{\int_{-\infty}^{\infty} e^{-\beta(\alpha x^2 - b x^3 - c x^4)} dx}$$

$$(1 + b x^3 + c x^4)$$

$\langle x \rangle = \frac{3b(K_B T)}{4a^2}$
--

Coeff. of thermal expansion:

Linear ^{or} expansion coeff.

$$\alpha = \frac{d\langle x \rangle}{dt} = \frac{3}{4a^2} b k_B = \text{const}$$

but expansion

$$\alpha \rightarrow 0 \text{ at } T \rightarrow 0$$

Quantum anharmonic osc. treatment:

$$\langle x \rangle = \frac{3}{4a^2} b \langle E \rangle$$

$$\langle x \rangle = \frac{3}{4a^2} b \left(\frac{\hbar\omega}{e^{\beta\hbar\omega} - 1} \right)$$

$$\alpha = \frac{d\langle x \rangle}{dT}$$

$$= \frac{3}{4a^2} b \left[\frac{(\hbar\omega)^2}{k_B T^2} \frac{e^{\hbar\omega/k_B T}}{(e^{\hbar\omega/k_B T} - 1)^2} \right]$$

$$\text{AT } T \rightarrow 0$$

$$\alpha \rightarrow 0$$

Q. Consider a system of two Ising spins s_1, s_2 taking values ± 1 with internal energy $E = -J s_1 s_2$. If it's in thermal equilibrium at temp. T , the ave. en. varies as $C/k_B T$ for large T .

then C varies as $-J^2, J^2, -J, +J$

$$\langle E \rangle = -\frac{\partial \ln Z}{\partial \beta}$$

$$Z = \sum g e^{-\beta E} = 2e^{-\beta J} + 2e^{\beta J}$$

$$E = +J$$

	s_1	s_2
$+J$	$+1$	-1
$-J$	-1	$+1$

$$\langle E \rangle = -\frac{[-2J e^{-\beta J} + 2J e^{\beta J}]}{2e^{-\beta J} + 2e^{\beta J}}$$

$$= \frac{J [e^{-J/K_B T} + e^{J/K_B T}]}{e^{-J/K_B T} + e^{J/K_B T}}$$

$$\langle E \rangle = \frac{J \left[1 - \frac{J}{K_B T} - 1 - \frac{J}{K_B T} \right]}{2}$$

$$\langle E \rangle = -\frac{J^2}{K_B T}$$

Q. The partition funⁿ of a system is $Z = e^{\alpha T^3 V}$, find N, P, S, U

α is constt. $H = 0$

$$P = \alpha K_B T^4, \quad S = 4\alpha K_B T^3 V, \quad U = 3\alpha K_B T^4 V$$

Q. The free en. of a gas having N particle, volume V & temp. T is given as

$$F = -N K_B T \ln \left[\frac{a_0 V (K_B T)^{5/2}}{N} \right]$$

$$\ln Z = \ln \left[\frac{a_0 V}{N \beta^{5/2}} \right]$$

$$E = -\frac{\partial}{\partial \beta} \ln Z$$

find internal energy of the gas. $= \frac{5}{2} \frac{N}{\beta}$

Q. The entropy $S = N K \ln \left[\frac{V}{h^3} \left(\frac{4\pi m E}{3N} \right)^{3/2} \right] + \frac{3}{2} N K_B$ in MCIE

Obtain.

$$(i) I.E. \rightarrow \frac{3}{2} N K_B T$$

Sacur-Tetrode eqn

$$(ii) C_V$$

$$(iii) P \Rightarrow \frac{P}{T} = \frac{\partial S}{\partial V} \Big|_{N, T}$$

$$(iv) \sigma = \frac{S}{K_B}$$

statistical error

$\tan h x \approx x$
 $x \approx 0$
 $x \approx 0$
 $\cos \approx 1$
 $\sin \approx x$

for system of N-particle $\Gamma = -Nk_B \ln(Z^N)$

$$\ln Z = \ln Z_1^N$$

$$= \ln \left[\frac{q_0 V}{N \beta^{5/2}} \right]^N$$

Q. The helmholtz energy F in case of canonical ensemble is given as

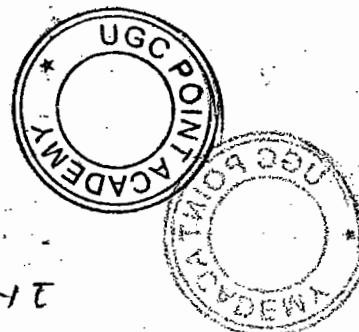
$$F = -NT \ln \left(\frac{eV}{Nt^2} \right)$$

$$t = \frac{\hbar}{\sqrt{2\pi m k}}, T = k_B T$$

find $\sigma \propto V(I \cdot E)$

$$\sigma = -\frac{\partial F}{\partial T} = N \ln \left(\frac{eV}{Nt^2} \right)$$

$$\sigma = -T^2 \frac{\partial}{\partial T} \left(\frac{F}{T} \right)_{V,N} = \frac{3}{2} NT$$



The entropy of mixing of two ideal gases
 (Gibbs paradox)

for a molecule of ideal gas

$$E = \frac{p_x^2}{2m} + \frac{p_y^2}{2m} + \frac{p_z^2}{2m}$$

$$Z_1 = \frac{V}{T^3} = V \left(\frac{2\pi mk_B T}{h^2} \right)^{3/2}$$

for N molecules

$$\begin{aligned} Z_{\text{system}} &= (Z_1)^N \\ &= V^N \left(\frac{2\pi mk_B T}{h^2} \right)^{3N/2} \\ &= V^N \left(\frac{2\pi m}{B h^2} \right)^{3N/2} \end{aligned}$$

$$\bar{E} = -\frac{\partial \ln Z}{\partial \beta}$$

$$\bar{E} = \frac{3}{2} N k_B T$$

$$\bar{p} = \frac{2}{3} \frac{\bar{E}}{V} = \frac{N k_B T}{V}$$

$$S = k_B \ln Z + \frac{\langle E \rangle}{T}$$

$$S = k \ln V^N + \frac{3}{2} N k \ln \left(\frac{2\pi m k_B T}{h^2} \right) + \frac{3}{2} N k_B$$

Now we consider the mixing of two ideal gases

T, V, N_1	T, V, N_2
$\vdots \vdots \vdots$	$\vdots \vdots \vdots$

ideal gas molecule
 is non-interacting
 so entropy remain same
 after \rightarrow before.

Before mixing the entropy of the system

$$S = S_1 + S_2 = 2S_1$$

$$S = 2Nk_B \ln V + 3Nk_B \ln \left(\frac{2\pi m k_B T}{h^2} \right) + 3Nk_B$$

After mixing the entropy of the system

$$N \rightarrow 2N, V \rightarrow 2V, T \rightarrow K$$

$$S' = 2Nk_B \ln 2V + 3Nk_B \ln \left(\frac{2\pi m k_B T}{h^2} \right) + 3Nk_B$$

Change in entropy

$$\Delta S = S' - S$$

$$\boxed{\Delta S = 2Nk_B \ln 2}$$

→ Mixing of two same ideal gases is reversible process

$$\Rightarrow \Delta S = 0$$

Two results are contradictory
particle is distinguishable.

This contrary result is known as Gibbs paradox.

* How to treat the problem arise in the result
→ modifying the partition function

$$Z = \frac{(Z_1)^N}{N!} \quad (\text{particle is indistinguishable})$$

$$Z = \frac{V^N}{N!} \left(\frac{2\pi m k_B T}{h^2} \right)^{3N/2}$$

Entropy before mixing

$$S = 2Nk_B \ln V - 2k_B \ln N! + 3Nk_B \ln \left(\frac{2\pi m k_B T}{h^2} \right) + 3Nk_B$$

$$\frac{1}{2} S_1$$

After mixing

$$S' = S$$

$$S' = N k_B \ln V - k_B \ln \underline{2N} + \frac{3}{2} N k_B \ln \left(\frac{2\pi m k_B T}{h^2} \right) + \frac{3}{2} N k_B$$

$$V \rightarrow 2V, N \rightarrow 2N, T \rightarrow K$$

$$S' = 2N k_B \ln 2V - k_B \ln \underline{2N} + 3N k_B \ln \left(\frac{2\pi m k_B T}{h^2} \right) + 3N k_B$$

$$\Delta S = S' - S$$



$$= 2N k_B \ln 2V - 2N k_B \ln V - k_B \ln \underline{2N} + 2k_B \ln \underline{LN}$$

$$= 2N k_B \ln 2 - [2N k_B \ln \underline{2N} - 2N k_B] + 2N k_B \ln N - 2N k_B$$

$$\Delta S = 0$$

→ Thus the Gibbs paradox is removed by taking into account the indistinguishable of particles. The modifying is correct expression of $Z(N, V, T)$ idealgas

$$Z(N, V, T) = \frac{V^N}{N!} \cdot \left(\frac{2\pi m k_B T}{h^2} \right)^{\frac{3N}{2}}$$

$$E = -\frac{\partial \ln Z}{\partial \beta} = \frac{3}{2} N k_B T$$

$$P = \frac{2}{3} \left(\frac{E}{V} \right)$$

$$S = N k_B \ln \left[\left(\frac{V}{N} \right) \left(\frac{4\pi m E}{3N} \right)^{3/2} \right] + \frac{5}{2} N k_B$$

which is called Sackur-Tetrode eqⁿ.

Maxwell's distribution law or molecular speeds ($v \rightarrow$)

Used Boltzmann's distribution law.

$dN(v)dv \Rightarrow$ No. of molecules in the range

$$\frac{dc}{c(c+dc)}$$

$$dN(c)dc$$

$$p \quad p+dp$$

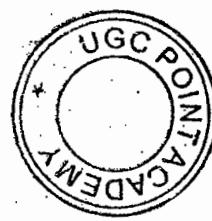
$$e \quad e+de$$

$$K \quad K+dK$$

If gas contain N molecules \rightarrow momentum

$$\frac{dN}{N} = dP = \frac{g(p) dp e^{-\beta E}}{Z}$$

Probab. No. of microstate
or degeneracy



$$= A g(p) dp e^{-\beta E}$$

$$= A \frac{\sqrt{4\pi p^2} dp e^{-\beta E}}{h^3}$$

$$dp = \frac{AVp^2 e^{-\beta(p^2/2m)}}{2\pi^2 h^3} dp$$

$$\int dp = \frac{AV}{2\pi^2 h^3} \int_0^\infty e^{-\beta(p^2/2m)} p^2 dp$$

$$= \frac{AV}{2\pi^2 h^3} \sqrt{\frac{2\pi}{(\beta/4m)^3}}$$

$$A = \frac{(2\pi)^{3/2} h^3}{V} \left(\frac{\beta}{4m} \right)^{3/2}$$

$$dP = \frac{4}{\sqrt{\pi}} \left(\frac{2}{2m k_B T} \right)^{3/2} p^2 e^{-\beta b^2/2m} dp$$

$p=mc$

$$\boxed{dP = \frac{dN}{N} = 4\pi \left(\frac{m}{2\pi k_B T} \right)^{3/2} e^{-mc^2/2k_B T} c^2 dc}$$

$$\boxed{dN = 4\pi N \left(\frac{m}{2\pi k_B T} \right)^{3/2} e^{-mc^2/2k_B T} c^2 dc}$$

$c \rightarrow 0 \text{ to } \infty$

Average speed : (\bar{c})



$$\bar{c} = \frac{\int c dP}{\int dP}$$

$$\int dP = 1$$

$$\bar{c} = \int c dP$$

$$= \int_0^\infty c \frac{dN}{N} = \frac{\int c dN}{\int dN}$$

$$\boxed{\bar{c} = \sqrt{\frac{8k_B T}{\pi m}}}$$

Avergespeed

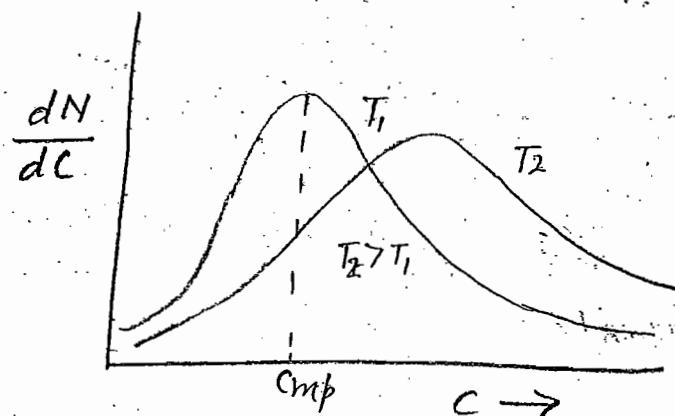
$$\bar{c^2} = \frac{\int c^2 dN}{\int dN}$$

$$= \frac{3k_B T}{m}$$

$$\left(\frac{C^2}{C^2}\right)^{1/2} = \sqrt{\frac{3k_B T}{m}} = C_{rms} \text{ root mean square}$$

$C_{rms} > \bar{C}_{ave} > C_{mp}$ (most probable)

$$\sqrt{3} : \sqrt{\frac{8}{\pi}} : \sqrt{2}$$



$$A_1 = A_2 \text{ Carea,}$$

$$\int \frac{dN}{dc} dc = N = \text{Breadth} \times \text{height}$$

Most probable speed (C_{mp}):

followed by max. no. of molecules

at $C = C_{mp}$

$$\frac{d}{dc} \left(\frac{dN}{dc} \right) = 0$$

$$\rightarrow \frac{dN}{dc} = 4\pi N \left(\frac{m}{2\pi k_B T} \right)^{3/2} c^2 e^{-mc^2/2k_B T}$$

peak depend upon temp.

Maxwell's distribution law of velocity distribution

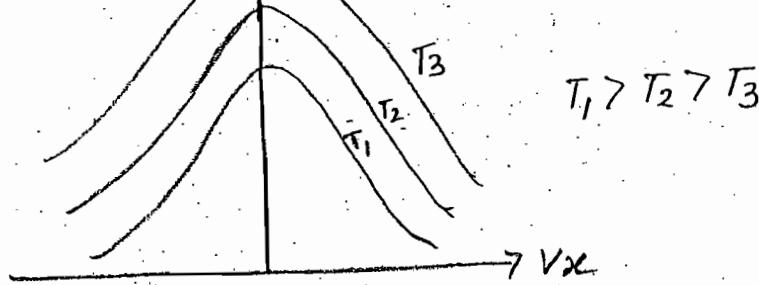
$$\frac{dN}{v_x \quad v_x + dv_x}$$

$$dP = \frac{dN(v_x)}{N} = \left(\frac{m}{2\pi k_B T} \right)^{1/2} e^{-mv_x^2/2k_B T} dv_x$$

$$dP(v) = \frac{dN(v)}{N} = \frac{dN(v_x) dN(v_y) dN(v_z)}{N}$$

$$= \left(\frac{m}{2\pi k_B T} \right)^{3/2} e^{-mv^2/2k_B T} \frac{dv_x dv_y dv_z}{dv}$$

$$\frac{dP}{dv_x} = \left(\frac{m}{2\pi k_B T} \right)^{1/2} e^{-mv_x^2/2k_B T}$$



i) $\frac{dP(v_x)}{dv_x} = \text{max. at } v_x = 0$

ii) Probability distribution curve is symmetric
 $P(-v_x) = P(v_x)$

Most Probable velocity

$$\Rightarrow \frac{d}{dv_x} \left(\frac{dP}{dv_x} \right) = 0 \text{ wrong}$$

$v_x = 0$ right

or $\frac{d}{dv_x} [P(v_x)] = 0$

Max. value of

$$\times \frac{dP(v_x)}{dv_x} \Big|_{max} = \frac{dP(v_x)}{dv_x} \Big|_{v_x=0} = \left(\frac{m}{2\pi k_B T} \right)^{1/2}$$

or

$$P(v_x) \Big|_{max} = P(v_x) \Big|_{v_x=0} = \left(\frac{m}{2\pi k_B T} \right)^{1/2}$$

i.v) Average velocity:

$$\langle v_x \rangle = \frac{1}{2} \int_{-\infty}^{\infty} v_x P(v_x) dv_x$$

$$= 0$$

v) $\langle v_x^2 \rangle$

$$= \int_{-\infty}^{\infty} v_x^2 P(v_x) dv_x$$

$$= \frac{k_B T}{m}$$

vi)

$$\langle v^2 \rangle = \langle v_x^2 \rangle + \langle v_y^2 \rangle + \langle v_z^2 \rangle$$

$$= \frac{k_B T}{m} + \frac{k_B T}{m} + \frac{k_B T}{m}$$

$$V_{rms} = \sqrt{\frac{3k_B T}{m}}$$

vii) Velocity at which $P(v_x)$ falls $1/e$ time of max.
value =

$$v_x = \sqrt{\frac{2k_B T}{m}}$$



$$\text{viii) } \overline{v_x v_y} = \iiint_{-\infty}^{\infty} v_x v_y p(v_x v_y) dv_x dv_y = 0$$

$$\text{ix) } \overline{v_x^2 v_y v_z} = 0$$

$$\text{x) } \overline{(\alpha + \beta v_x)^2} = \alpha^2 + \beta^2 \frac{k_B T}{m}$$

$$\text{x i) } \overline{(\alpha v_x - \beta v_y)^2} = (\alpha^2 + \beta^2) \frac{k_B T}{m}$$

$$\text{x ii) } \overline{(\alpha v_x + \beta v_y - v_z)^2} = (\alpha^2 + \beta^2 + 1) \frac{k_B T}{m}$$



Quantum Statistical Mechanics :

→ Applicable when the particles are identical & indistinguishable.

Fermi-Dirac statistics
Particle have spin
 $s = \frac{1}{2}, \frac{3}{2}, \dots$
(half odd integral spin)

Ex: e^- , p , n etc. π^- , $Hole$
These particles are Klas fermions.

→ fermions follow the pauli exclusion principle.

→ Condensation property not exist.

Bose-Einstein st.

Particle have spin

$s = 0, 1, 2, \dots$ (integral)

Ex: ${}^2He^4$, 1H atom, ${}^3Li^7$ etc
~~alpha~~, positron!
These particles are Klas.

Bosons

→ Bose-Eins condensation is possible.

→ They do not follow P.E.Y.,
→ No. of distribution is greater than fermions.

→ In case of Boson

${}^{11}Na^{23}$

In case of atom = $n + p + e$

$$= A - Z + Z + Z$$

$$= A + Z \rightarrow \text{even} \rightarrow \text{Boson}$$

Nucleus

$$A - Z + Z = A \rightarrow \text{even}$$

Ex: ${}^{11}Na^{23}$ ${}^{11}Na^{24}$
Bosons fermions

a b
q. states

$$\psi_{ab}^{MB}(12) = \phi_a(1) \phi_b(2)$$

$$\psi_{ab}^{(1,2)} = \frac{1}{\sqrt{2}} [\phi_a(1) \phi_b(2) - \phi_a(2) \phi_b(1)]$$

F-D sto wave funⁿ is antisymmetric.

$$\psi_a^f(1,2) = 0, \text{ when } a \equiv b$$



$$= \frac{1}{\sqrt{2}} \begin{vmatrix} \phi_a(1) & \phi_a(2) \\ \phi_b(1) & \phi_b(2) \end{vmatrix}$$

Slater Determinant.

for Bosons:

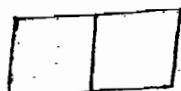
$$\psi_{ab}^B(1,2) = \frac{1}{\sqrt{2}} [\phi_a(1) \phi_b(2) + \phi_a(2) \phi_b(1)]$$

→ wave funⁿ is symmetric.

$$a \equiv b$$

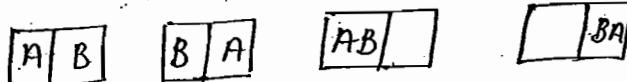
$$\underset{a \equiv b}{\psi_{ab}^B(1,2)} = \sqrt{2} \phi_a(1) \phi_a(2)$$

Distribution of two particles in two degenerate cells →



case-1) M-B/classical :

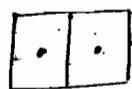
$$N = g_i n_i$$



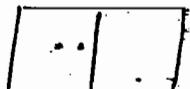
2.) Bosons:

No. of ways of distribution

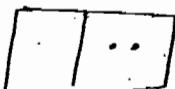
$$\Omega = \frac{n!}{\prod_i g_i!} c_{ni}$$



or



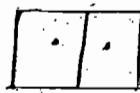
or



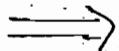
$$\Omega_{B-E} = \frac{2+2-1}{2} C_2 = 3 C_2 = 3$$



3.) Fermions:



$$\Omega_{F-D} = g_i c_{ni}$$



No. of ways of distribution of N particles in different cells $1, 2, \dots, i$ having degeneracies g_1, g_2, \dots etc.

(i) when particles are classical:

$$\Omega_{m-B} = \prod_i \frac{\text{LN}}{\text{L} n_i} (g_i)^{n_i}$$



2.) Bosons:

$$\Omega_B = \prod_i \frac{n!}{\prod_i g_i!} c_{ni}$$

3.) Fermions:

$$\Omega_{F-D} = \prod_i (g_i c_{ni})$$

~~gmr~~ ratio of probabilities of finding the two particles in some state to that probability of finding particles in different states.

$$r_{MB} : r_{BE} : r_{FD} = 1 : 2 : 0$$

Q. Partition fun" for a system of two particles each of which can occupy any one of energy level o se.

Soln:

(i) If particle are classical:

Distr.		En.	
AB	0	ϵ	ϵ
A	B	ϵ	ϵ
B	A	ϵ	ϵ

$Z = \sum_i g_i e^{-\beta E_i}$
 $= g_1 e^{-\beta E_1} + g_2 e^{-\beta E_2} + g_3 e^{-\beta E_3}$
 $= 1 \cdot e^0 + 2 e^{-\beta \epsilon} + e^{-2\beta \epsilon}$

$$\boxed{Z = (1 + e^{-\beta \epsilon})^2}$$

(ii) fermions:

0	ϵ	En
•	•	ϵ

$$Z = e^{-\beta \epsilon}$$

iii) Bosons:

乙

6

6

2 €

$$Z = e^{-\beta E} + e^{-0} + e^{-2\beta E}$$

$$Z = 1 + e^{-\beta E} + e^{-2\beta E}$$



Q. Two particles are to be distributed in different cells having energies $0, \epsilon, 2\epsilon, \dots$

The ground state is non-degenerate, while the excited states are doubly degenerate.

find the ways of distribution such that the E_n of the system is 3E if particles are:

Sol^x

(i) fermions:

36

۲۶

2

6

9

2 - 4

Ways of distribution = 6

$$g_i c_{n_i} = 4c_2 = \frac{14}{12 \cdot 12} = \frac{4x^2 \cdot 3 \times 5}{12 \cdot 12} = 6$$

ii) Bosons:

~~ask~~

$$\text{No. of ways} = 6$$

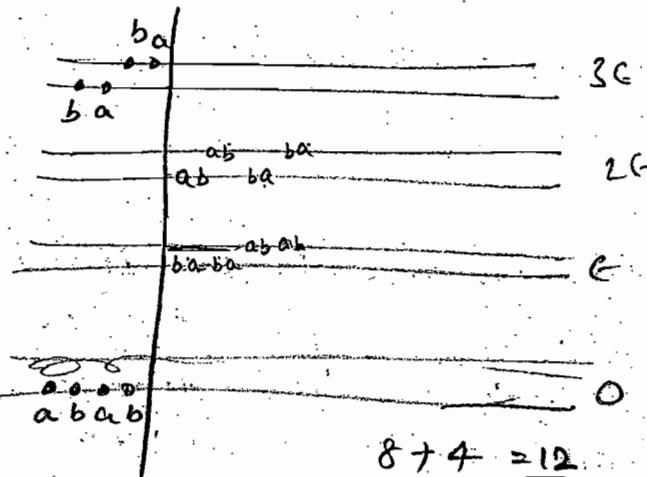
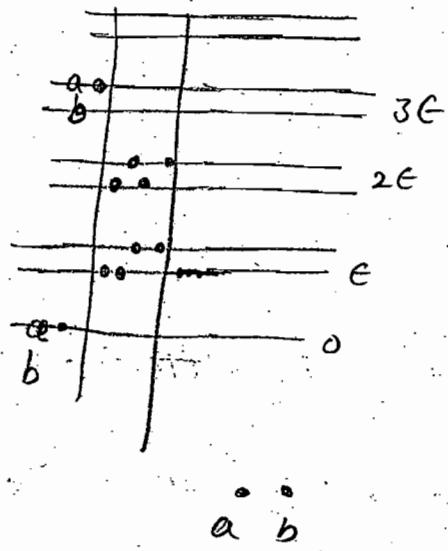
$$n_i + g_i - 1 \leq n_i^{(2+4-1/c_2)}$$

$$\frac{5}{15} C_2 = \frac{5 \times 4}{12} \frac{2}{12}$$

iii) classical:

$$\text{No. of ways} = 12$$

$$\underline{3^2 = 9}$$



$$8 + 4 = \underline{\underline{12}}$$



Fermi-Dirac statistics:

Distribution funⁿ for fermions $f(\epsilon) \rightarrow$

We KLT, for M-B & to

$$f(\epsilon_i) = \frac{n_i}{g_i} = \frac{1}{e^{\epsilon_i + \beta E_i}} = \frac{1}{e^{(\epsilon_i - \mu)/k_B T}}$$

In F-D & to

$$\Omega_{F-D} = g_i c_{ni}$$

$$= \frac{g_i}{\underline{n_i} \underline{g_i - n_i}}$$

At equil^m

$$d(\log \Omega_{F-D}) = 0$$

$$N - \sum_i n_i = 0 \quad \left. \right\} \text{constraints eq's}$$

$$E - \sum_i n_i \epsilon_i = 0$$

$$f(\epsilon_i) = \frac{n_i}{g_i} = \langle n_i \rangle$$

$$= \frac{1}{e^{\beta(\epsilon_i - \mu)} + 1}$$

2nd method:

Partition funⁿ

$$Z = \sum_i e^{-\beta(\epsilon_i - \mu)}$$



$$Z = \sum_{n_1, n_2, \dots} e^{-\beta E_1 n_1 - \beta E_2 n_2 - \dots} = e^{-\beta E_1} e^{-\beta E_2} \dots$$

or

$$Z = \sum_{n_1=0}^{\infty} e^{-n_1 \beta (E_1 - H)} \sum_{n_2} e^{-n_2 \beta (E_2 - H)}$$

or

$$Z = (1 + e^{-\beta (E_1 - H)}) \times (1 + e^{-\beta (E_2 - H)})$$

$$Z = \prod_i (1 + e^{-\beta (E_i - H)})$$

$$\log_e Z = \sum_i \log (1 + e^{-\beta (E_i - H)})$$

$$\sum_i \langle n_i \rangle = N = \frac{1}{\beta} \frac{\partial}{\partial H} (\log_e Z) = \frac{1}{\beta} \sum_i \frac{\beta e^{-\beta (E_i - H)}}{(1 + e^{-\beta (E_i - H)})}$$

$$TdS = dE + PdV - HdN + NdH$$

$$N = -\frac{\partial E}{\partial H} / \left(\frac{1}{\beta} \frac{\partial \ln Z}{\partial H} \right)$$

$$N = -\frac{1}{\beta} \frac{\partial}{\partial H} \ln Z |_{T, V}$$

$$\boxed{\langle n_i \rangle = \frac{1}{e^{\beta (E_i - H)} + 1}}$$

Ave. no. of occupation no.
particle in i th
cells

Occupation probability:

$$f(E) = \frac{1}{e^{\beta (E - H)} + 1}$$

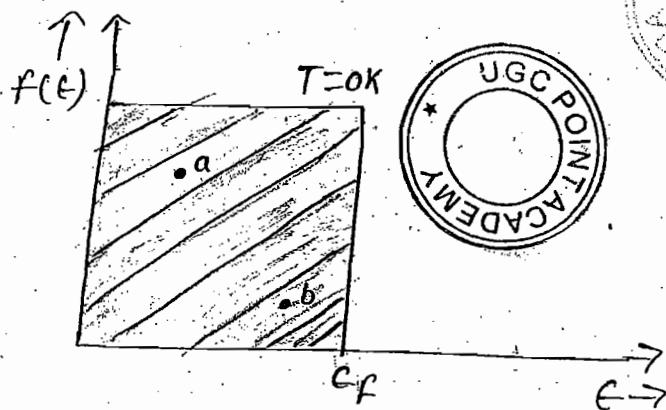
$$f(\epsilon) = \frac{1}{e^{(\epsilon - E_F)/k_B T} + 1}$$

$$\beta = \frac{1}{k_B T} \quad k_B T \sim E_F$$

at T = 0K

$$f(\epsilon) = 1 \quad \epsilon < E_F$$

$$= 0 \quad \epsilon > E_F$$



Expression for

$E_F(T=0K)$ for fermions or fermi gas at 0K →

No. of states in ϵ -range

$$\epsilon \rightarrow \epsilon + d\epsilon$$

$$\beta \rightarrow \beta + d\beta$$

$$g(\beta) d\beta = \frac{V 4\pi \beta^2 d\beta}{h^3}$$

$$g(\epsilon) d\epsilon = \frac{V 4\sqrt{2} \pi m^{3/2} E^{1/2} dE}{h^3}$$

No. of e- (fermions) $\xrightarrow{\text{at } 0^\circ K}$

$$dn(\epsilon) = (g(\epsilon) d\epsilon \times f(\epsilon)) \times g_s \xrightarrow{(2s+1)} \text{Fermions}$$

No. of filled states

Total no. of fermions

$$N = \int d\tau n(\epsilon) = \int dn \epsilon(\epsilon)$$

$$= (2s+1)V \frac{4\sqrt{2} \pi m^{3/2}}{h^3} \int_0^{E_f(0)} f(\epsilon) \epsilon^{1/2} d\epsilon$$

↓
↑ If T=0K



$$N = (2s+1) \frac{4\sqrt{2} \pi m^{3/2}}{h^3} \frac{E_f(0)^{3/2}}{3/2}$$

$$E_f(0) = \frac{\hbar^2}{2m} \left[\frac{3N}{4\pi V g_s} \right]^{2/3}$$

$$E_f(0) \propto \left(\frac{N}{V} \right)^{2/3}$$

$$E_f(0) \propto n^{2/3}$$

$n \rightarrow$ fermion density

Average energy of fermion in Fermi-Dirac gas \rightarrow

$$\bar{\epsilon} = \frac{\int_0^{E_f(0)} \epsilon dn}{\int_0^{E_f(0)} dn}$$

$$= \frac{\int_0^{E_f(0)} \epsilon \epsilon^{1/2} dn}{\int_0^{E_f(0)} \epsilon^{1/2} dn}$$

$$\bar{\epsilon} = \frac{3}{5} E_f(0)$$

Ave. En. of gas:

$$\bar{E}_{\text{gas}} = N \bar{E} = \frac{3}{5} N E_f(0)$$

At OK \rightarrow

$$E_f(0) = \frac{p_f^2}{2m} + \frac{\hbar^2 k_f^2}{2m}$$

$k_f \rightarrow$ fermi wave vect

$$= \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2)$$

$$k_f \propto n^{1/3}$$

$$p_f \propto n^{1/3}$$

$$v_f \propto n^{1/3}$$

(velocity)

$$t_f \propto n^{-1/3}$$

$$F_f \propto n^{2/3}$$

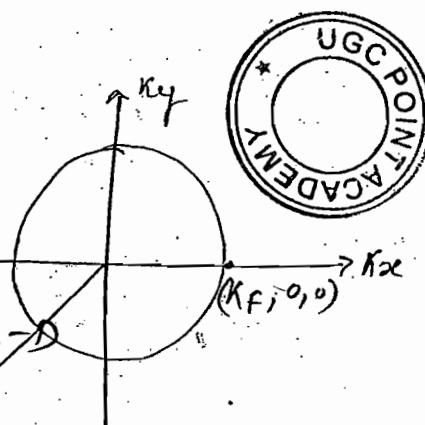
fermi temp \propto Energy

$$k_F = \left(\frac{3\pi^2 N}{V} \right)^{1/3} \text{ for 3-D}$$

$$k_F = \left(\frac{2\pi N}{A} \right)^{1/2} \rightarrow 2D \text{ fermi surface at}$$

OK \rightarrow spherical

$$k_F = \left(\frac{N\pi}{2a} \right)^{1/3} \rightarrow 1-D$$



Pressure of the gas:

$$= \frac{1}{3} g_s \left(\frac{\bar{E}_{\text{gas}}}{V} \right)$$

$$= \frac{1}{3} (2s+1) \frac{\frac{3}{5} N E_f(0)}{V}$$

$$= \frac{1}{5} (2s+1) \left(\frac{N}{V} \right) E_f(0) \quad \text{--- (1)}$$

$$PV = \frac{1}{3} g_s \bar{E}$$

$$PV = \frac{2}{3} \bar{E}$$

$$\bar{E} = a \left(\frac{N}{V} \right)^{5/3} \cdot \frac{1}{V^{2/3}} = a \frac{N^{1/3}}{V^{2/3}}$$

$$P = -\frac{\partial E}{\partial V} \Big|_N = \frac{2}{3} a (N)^{5/3} V^{-5/3}$$

$$PV = \frac{2}{3} \bar{E} = \frac{1}{3} g_s \bar{E}$$



from ①

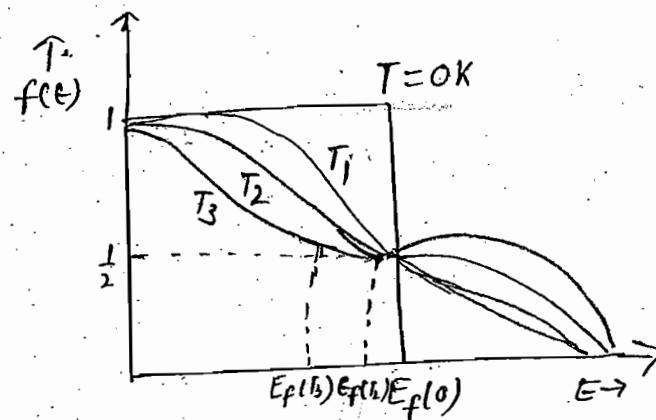
$$P = \frac{1}{3} \left(\frac{N}{V} \right)^{5/3} \left(\frac{6\pi^2}{2S+1} \right)^{2/3} \frac{t^2}{m}$$

$$P \propto N^{5/3}$$

At 0K

$$P_{MB} = 0$$

$$P_{FD} > P_{MB}$$



$$f(E) = \frac{1}{e^{(E-E_f)/k_B T} + 1}$$

$T \neq 0K$

$$E = E_f$$

$$f(E) = \frac{1}{2}$$

$$E_F(T) = E_F(0) \left[1 - \frac{5\pi^2}{h^2} \left(\frac{K_B T}{E_F(0)} \right)^2 \right]$$

→ Pressure increases when E_F increases.

Q. Show that at $T=0K$ in e^- -gas, where $\epsilon \ll E_F(0)$

$$\frac{N(E_F - \epsilon \rightarrow E_F)}{N_{\text{total}}} = \frac{3}{2} \frac{\epsilon}{E_F}$$

Soln:

$$\frac{\int_{E_F-\epsilon}^{E_F} dn}{\int_0^{E_F} dn} = \frac{\int_{E_F-\epsilon}^{E_F} E^{1/2} dE}{\int_0^{E_F} E^{1/2} dE}$$

$$= \left[1 - \left(\frac{E_F - \epsilon}{E_F} \right)^{3/2} \right]$$

$$= \left[1 - \left(1 - \frac{\epsilon}{E_F} \right)^{3/2} \right]$$

$$= \frac{3}{2} \frac{\epsilon}{E_F}$$



Q. for a 2-D e^- gas no. of e^- per unit area is given by

$$n = \frac{4\pi m k_B T}{h^2} \log \left(e^{E_F/k_B T} + 1 \right)$$

Soln:

$$\frac{N}{A} = n = \frac{4\pi m k_B T}{h^2} \log \left(e^{E_F/k_B T} + 1 \right)$$

$$\frac{dN}{A} = dn = g_s \times g(k) dK \times f$$

$$= 2 \times \frac{k}{2\pi} dK \times \frac{1}{e^{(E_F - EF)/k_B T} + 1}$$

$$= \frac{2m}{2\pi h^2} \int_0^{\infty} \frac{1}{e^{(E - E_F)/k_B T} + 1} dt$$

Q. Show that the de-Broglie wavelength of e⁻ at termi. surface at OK is
 $2\left(\frac{\pi}{3n}\right)^{1/2}$.

Hint: $\lambda_{dB} = \frac{h}{p_f} = \frac{h}{\sqrt{2mE_f}}$



Q. Show that the partition funⁿ of a relativistic gas of N monoatomic molecule having the energy momentum relation $E=pc$ is

$$\frac{1}{N!} \left[\frac{8\pi v}{h^3} \left(\frac{k_B T}{c} \right)^3 \right]^N$$

$$g(b)db = g(c)c dc$$

Also find $\langle E \rangle$, $\langle p \rangle$

Solⁿ:

$$\begin{aligned} Z &= \sum g_i e^{-\beta E_i} \\ &= \int g(e) d\epsilon e^{-\beta E_i} \\ &= \int g(b) db e^{-\beta pc} \\ &= \int_0^\infty \frac{4\pi b^2 db V}{h^3} e^{-\beta pc} \\ &= \frac{8\pi v}{h^3} \left(\frac{k_B T}{c} \right)^3 \end{aligned}$$

$$Z_{\text{gas}} = \frac{1}{N!} (z_1)^N$$

$$= \frac{1}{N!} \left[\frac{8\pi V}{h^3} \left(\frac{k_B T}{c} \right)^3 \right]^N$$

$$\langle E \rangle = - \frac{\partial \ln Z}{\partial \beta} \Big|_{V, N}$$

$$\langle P \rangle = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} \Big|_{T, N}$$

$$\langle P \rangle = \frac{1}{\beta} \frac{\langle E \rangle}{V}$$

→ If case is non-relativistic then $E = p^2/2m$ treated this prob

Q. Show that the partition funⁿ of a non-relativistic ga
of N monoatomic molecules having the eq. mom. reln

$$E = p^2/2m$$

$$z_1 = \sum g_i e^{-\beta E_i}$$

$$\langle P \rangle = \frac{2}{3} \frac{\langle E \rangle}{V}$$

$$z_1 = \frac{V}{73}$$

Q. Calculate the term energy & average en of e^- in ev for sodium assuming it has one free e^- per atom.

Given $P = 0.977 \text{ gm/cm}^3$ $A = 23$

Soln:

$$E_f = \frac{\hbar^2}{2m_e} \left(\frac{3N}{8\pi r} \right)^{2/3}$$

$$\frac{3}{5} E_0$$

$$\rho = \frac{M}{V} = \frac{N \cdot m}{V}$$



Q. 3 spin $1/2$ fermions are distributed in 2 non-degenerate cells
no. of ways of distribution

		E'		4 ways
E		\uparrow	$\uparrow \downarrow$	
\uparrow	\downarrow	$\uparrow \downarrow$	$\uparrow \downarrow$	4 ways
	\uparrow	\uparrow	\downarrow	
$\uparrow \downarrow$	\uparrow	\uparrow	\downarrow	2 ways
	\downarrow	\downarrow	\uparrow	
		E'		5 ways
		$\uparrow \downarrow$	\downarrow	0 ways

Bose-Einstein statistics:

Distribution function:

$$Z = \sum_i e^{-\beta(E_i - H)}$$

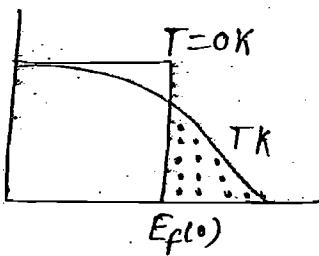
$$= \sum_{n_1, n_2} e^{-\beta[(n_1 E_1 + n_2 E_2 + \dots) - H(n_1 + n_2 + \dots)]}$$

$$= \sum_{n_1=0}^{\infty} e^{-\beta n_1 (E_1 - H)} \times \sum_{n_2} e^{-\beta n_2 (E_2 - H)} \quad \sum x^n = \frac{1}{1-x}$$

$$Z = \frac{1}{1-e^{-\beta(E_1 - H)}} \times \frac{1}{1-e^{-\beta(E_2 - H)}} \times \dots$$

$$Z = \prod_i \frac{1}{[1 - e^{-\beta(E_i - H)}]}$$

Electronic specific heat (c_e) \rightarrow



No. of excited es in range dE

$$N_{\text{excited}} = \frac{8\sqrt{2}V\pi m^{3/2}}{h^3} \frac{dE}{k_B T}$$

$$N_{\text{excited}} = \frac{3}{2} \frac{N k_B T}{E_f(0)}$$

Excitation Energy

$$E = N_{\text{exc}} \times k_B T$$

$$= \frac{3}{2} \frac{N k_B^2 T^2}{E_f(0)} = k_B T_F$$

$$c_e = \frac{dE}{dT} = \frac{3N k_B^2 T}{k_B T_F}$$

$$c_e = 3N k_B \left(\frac{T}{T_F}\right)$$

$$\boxed{c_e \propto T}$$

$$c_e = \alpha T$$

Bose-Einstein distribution functions

At page

$$Z = \prod_i \frac{1}{[1 - e^{-\beta(E_i - H)}]}$$

$$\ln Z = - \sum_i \ln [1 - e^{-\beta(E_i - H)}]$$

$$\sum_i \langle n_i \rangle = N = \frac{1}{\beta} \frac{\partial \ln Z}{\partial H} \Big|_{T, V}$$

$$\sum_i \langle n_i \rangle = \sum_i \frac{1}{e^{\beta(E_i - H)} - 1}$$

$$\boxed{\langle n_i \rangle = \frac{1}{e^{\beta(E_i - H)} - 1}}$$

Occupation no. for Bosons:

$$\langle n \rangle = \frac{1}{e^{\beta(E - H)} - 1}$$

$$f_{BE}(E) = \langle n \rangle = \frac{1}{e^{\beta E} \cdot e^{-\beta H} - 1}$$

$$e^{-\beta H} \gg 1$$

$$f_{BE} = f_{mB}$$

Q. st. change into cl. st.

Classical ideal gas:

$$H = -\frac{1}{\beta} \frac{\partial \ln Z}{\partial N} \Big|_{T, V}$$

$$Z = \frac{V^N}{h^{3N/2}} (2\pi m k_B T)^{3N/2}$$

$$= N! \ln ((2\pi m k_B T)^{3/2} + N \ln N - N \ln N + N)$$

$$M = -\frac{1}{\beta} \left[\ln \left\{ \frac{V}{N} \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2} \right\} \right]$$

for validity of CS'M

$$e^{-\beta E} \gg 1$$

$$= \frac{V}{N} \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2} \gg 1$$

$$\Rightarrow \frac{1}{n^{1/3}} \gg 1$$

$$[n^{1/3} \ll 1]$$

Validity cond'n for CS'M.

When T high

$$n = \frac{N}{V} = \text{low}$$

particle behave like cl. gas

when T is low

$$n = \frac{N}{V} = \text{High}, n^{1/3} \gg 1$$

particle behave like quantum gas particle.

Ideal Bose gas \rightarrow
(Photon gas)

for photons

$$Z_{photon} = \frac{1}{i} \frac{1}{[1 - e^{-\beta(E_i - \mu)}]}$$

With $\mu = 0$, (min en. of a particle or photon
 $E = mc^2 = 0$)

$$Z_{photon} = \frac{1}{i} \frac{1}{(1 - e^{-\beta E_i})}$$

$$\ln Z = \sum_i \ln \frac{1}{(1-e^{-\beta E_i})}$$

$$\ln Z_{\text{photongas}} = 2 \times \int g(b) db \quad \ln Z/\text{photon}$$



$$= 2 \times \int g(\omega) d\omega \ln \left(\frac{1}{1-e^{-\beta E}} \right)$$

$$g(b) db = \frac{4\pi b^2 db}{h^3} V = \frac{4\pi E^2 dE}{h^3 c^3} V$$

$$E = \hbar\omega$$

$$g(\omega) d\omega = \frac{V \times 4\pi \hbar^3 \omega^2 d\omega}{h^3 c^3}$$

$$\ln Z_{\text{ph. gas}} = - \int_0^\infty \frac{\omega^2 V}{\pi^2 c^3} \ln(1-e^{-\beta \hbar \omega}) d\omega$$

$$\boxed{\ln Z_{\text{ph. gas}} = \frac{V \pi^2}{45 c^3 (\beta \hbar)^3}}$$

$$F = -k_B T \ln Z = -k_B T \frac{V \pi^2}{45 c^3 (\beta \hbar)^3}$$

Av. En. of photon gas:

$$\langle E \rangle = -\frac{\partial \ln Z}{\partial \beta} |_{N,V}$$

$$= \frac{3V \pi^2}{45 c^3 \hbar^3} \left(\frac{1}{\beta^4} \right)$$

$$\boxed{\langle E \rangle \propto T^4}$$

w freq find 2 mode
Transvers mode
for photon
 $\omega \sim \omega_0 \text{ dyn}$

$$\langle P \rangle = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} \Big|_{T, N}$$

$$= \frac{\pi^2}{45 c^3 \hbar^3 \beta^4}$$

$$\boxed{\langle P \rangle \propto T^4}$$

$$\rightarrow \frac{\langle P \rangle}{\langle E \rangle} = \frac{1}{3V}$$

$$\Rightarrow \langle P \rangle = \frac{1}{3} \frac{\langle E \rangle}{V} \rightarrow \text{Relativistic case}$$

$$\boxed{\langle P \rangle = \frac{\langle U \rangle}{3}}$$

$\langle U \rangle \rightarrow$ En. density

specific heat at constt volume

$$c_V = \frac{\partial \langle E \rangle}{\partial T}$$

$$= \frac{32 \pi^5 V k_B^4 F^3}{15 (\hbar c)^3}$$

$$\boxed{c_V \propto T^3}$$

*

$$\text{Entropy } S = \frac{-\partial F}{\partial T} \Big|_{N, V}$$

$$= \frac{32 \pi^5 k_B^4 V T^3}{45 (\hbar c)^3}$$

$$\boxed{S \propto T^3}$$



$$\frac{N \cdot R}{P V} = \frac{S}{3} \langle E \rangle$$

$$\langle P \rangle = \frac{2}{3} \frac{\langle E \rangle}{V}$$

Planck's radiation formula of BBR

B.B. has atomic oscillators
and radiation has discrete En. packets
(photons)



Energy of osc.

$$E_n = nh\nu$$

$$n = 0, 1, 2, \dots$$

Total no. of osc. $\propto e^{-h\nu/K_B T}$

$$N = N_0 + N_1 + N_2 + \dots = \frac{N_0}{1 - e^{-h\nu/K_B T}}$$

Total En. of osc.

$$E = E_0 N_0 + E_1 N_1 + \dots = \frac{N_0 h\nu e^{-h\nu/K_B T}}{(1 - e^{-h\nu/K_B T})^2}$$

Ave. En. of an osc.

$$= \frac{E}{N}$$

$$= \frac{h\nu}{e^{h\nu/K_B T} - 1} = \frac{h\nu}{e^{h\nu/K_B T} - 1}$$

No. of oscillators in freq. range $\nu \rightarrow \nu + d\nu$

$$= 2 \times \frac{4\pi b^2 d\nu}{h^3}$$

Energy density of black body having freq. $\nu \rightarrow \nu + d\nu$

$$u_\nu d\nu = \frac{\text{no. of osc. in range } \nu \rightarrow \nu + d\nu}{\text{volume}} \times \bar{E}_{\text{osc}}$$

$$u_\nu d\nu = \frac{8\pi h\nu^3}{c^3} \frac{1}{e^{h\nu/K_B T} - 1} d\nu$$

Planck's radiation formula

$$p = E/c = h\nu/c$$

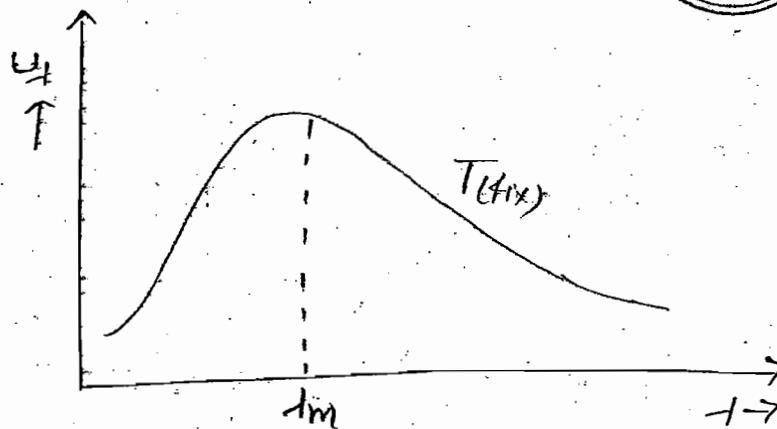
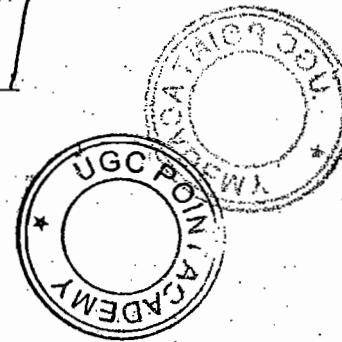
In terms of wavelength

$$\nu = c/\lambda$$

$$d\nu = -\frac{c}{\lambda^2} d\lambda$$

$$U_1 d\lambda = \frac{8\pi h c}{\lambda^5} \frac{1}{e^{hc/\lambda k_B T} - 1} d\lambda$$

$$U_1 = \frac{8\pi h c}{\lambda^5} \frac{1}{e^{hc/\lambda k_B T} - 1}$$



→ BB spectra is continuous. (i.e. it absorb completely, completely, entirely)

(i) In low wavelength limit

$$\frac{hc}{k_B T} \gg 1$$

$$U_1 d\lambda = \frac{8\pi h c}{\lambda^5} \times \cancel{\left(e^{-hc/\lambda k_B T} \right)}$$

$$U_1 d\lambda = \frac{8\pi h c}{\lambda^5} e^{-hc/\lambda k_B T} d\lambda$$

This is Wein's Law.

(ii) long wavelength limit

$$\frac{hc}{k_B T} \ll 1 \Rightarrow e^{hc/k_B T} = 1 + \frac{hc}{k_B T}$$

$$u_d dt = \frac{8\pi k_B T}{14} dt$$

Rayleigh's - Jean's law.

(iii) u_d max. at $\lambda = \lambda_m$

$$\Rightarrow \frac{du_d}{d\lambda} = 0$$

$$\Rightarrow \frac{3(e^{hc/\lambda_m k_B T} - 1)}{e^{hc/\lambda_m k_B T}} = \frac{hc}{\lambda_m k_B T}$$

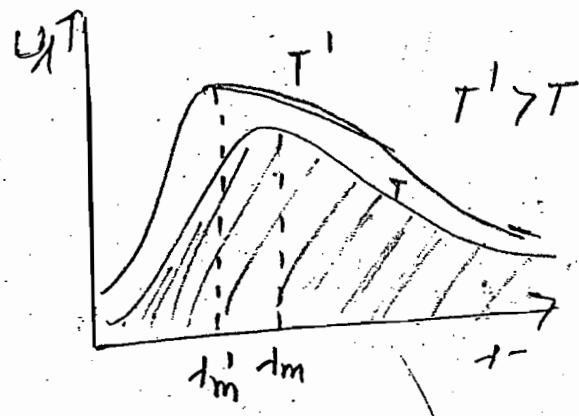
$$\Rightarrow 3 = \frac{hc}{\lambda_m k_B T}$$

$$\Rightarrow \lambda_m \cdot T = \frac{hc}{3k_B} = \text{const.}$$

$$\lambda_m \cdot T = \frac{hc}{3 \cdot 965 k_B} = b$$

Wein's displacement law

$$\lambda_m T' = \lambda_m T$$



(iv) Area enclosed by the curve
 $\propto T^4$
 (Stefan's Law)

$$\begin{aligned} \text{Area} &= \int_0^\infty v_T dt \\ &= \int_0^\infty \frac{8\pi hc}{15} \frac{1}{e^{hc/kT} - 1} dt \\ &\quad \frac{hc}{kT} = x \\ &= \propto T^4 \int_0^\infty x^3 \frac{dx}{e^x - 1} \end{aligned}$$



Stefan's Law:

$$E = \sigma A e(T^4 - T_0^4) t$$

$e = 1$ for B.B.

$$\sigma = 5.67 \times 10^{-8} \text{ watt/m}^2$$

$$\sigma = \frac{2\pi^5 k_B^4}{15 h^3 c^2}$$

Q. $T_0 = 27^\circ\text{C}$, $T_A^{eq} = 373^\circ\text{C}$, $T_B^{eq} = 573^\circ\text{C}$. Body A & B are spherical of radius $\frac{r_A}{r_B} = \frac{1}{2}$. find

$$\frac{\text{Rate of cooling of body A}}{\text{Rate of cooling of body B}} = \frac{1}{4} \left(\frac{T_A^4 - T_0^4}{T_B^4 - T_0^4} \right)$$

\Rightarrow

$$\text{Rate of cooling} \propto (T - T_0)$$

if $(T - T_0)$ is small

This is Newton's Law of cooling.



Adiabatic expansion of black body \Rightarrow

$$dQ = dU + PdV$$

$$\begin{aligned} \text{In adiab. } & dQ = dU + PdV \\ &= d(UV) + PdV \\ &= udv + vdu + PdV \\ &= vdv + vdu + \frac{2}{3}dv \end{aligned}$$

$$\begin{aligned} \Rightarrow 0 &= \frac{4u}{3}dv + vdu \\ &= \frac{4}{3} \frac{dv}{v} + \frac{du}{u} \end{aligned}$$

$$UV^{4/3} = \text{constt}$$

$$u \propto T^4$$

$$T^4 V^{4/3} = \text{constt} \rightarrow P^{1/4} V^{1/3} = \text{constt}$$

$$PV^{4/3} = \text{constt}$$

$$TV^{1/3} = \text{constt}$$

$$or VT^3 = \text{constt}$$

$$TV^{1/3} = \text{constt}$$

$$\gamma = \frac{C_P}{C_V} = \frac{4}{3} \rightarrow \text{Triatomic gas}$$

If adiabatically in case of BB:

$$T \rightarrow 2T$$

$$V \rightarrow \frac{V}{8}$$

Measurement of Sun's temp \rightarrow

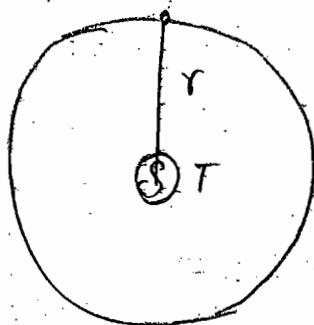
By pyrometer

Total radiation
pyrometer

(Based on Stefan's
law)

optical pyrometer

(Based on Be Wein's displacements)



En. radiated by the Sun/ sec

$$= \sigma \cdot 4\pi R_s^2 T^4$$

$$= S \times 4\pi r^2$$

\downarrow
En. falling/Area . sec on earth's
surface. (solar constt)

(Poynting vector) unit w/m^2

$$S = 2 \text{ cal/min-cm}^2$$

BOSE-EINSTEIN CONDENSATION:

particle-density \rightarrow sharp increase \Rightarrow condensation.

Occupation No.

$$N = \sum_s \langle n_s \rangle = \sum_s \frac{1}{e^{B(E_s - \mu)} - 1}$$

$$\langle n \rangle = \frac{1}{e^{B(E - \mu)} - 1}$$

for gr. state $E = 0$

$$\langle n \rangle = \frac{1}{e^{-\beta \mu} - 1}$$

for Bosons μ must be -ve.
as $\langle n \rangle$ can never be -ve.

$$\Rightarrow \langle n \rangle = \frac{1}{e^{\beta |\mu|} - 1}$$

$T \rightarrow$ Increases

$|\mu| \rightarrow$ Increases

$T \rightarrow$ Decreases

$|\mu| \rightarrow$ decreases.

As no. of particle is conserved.
The temp. at which
 $|\mu| \rightarrow 0$, when T is very low

is called $T \rightarrow T_B$ (Bose temp. or condensation temp.)

No. of Bosons in the Bose gas

$$N = \int g(p) dp \times f_{BE}(E)$$

$$g(p) dp = \frac{4\pi p^2 dp \cdot V}{h^3} = \frac{V p^2 dp}{2\pi^2 \hbar^3}$$

$$N = \frac{V}{2\pi^2 h^3} \int_0^\infty \frac{p dp}{e^{\beta(p^2/2m - U)} - 1}$$

at $T = T_B$, $U = 0$

$$N = \frac{V}{2\pi^2 h^3} \int_0^\infty \frac{p^2 dp}{e^{p^2/2mK_B T} - 1}$$

$$\text{let } y^2 = \frac{p^2}{2mK_B T}$$

$$2y dy = \frac{2p dp}{2mK_B T_B}$$

$$dp = \sqrt{2mK_B T_B} dy$$

$$N = \frac{V(2mK_B T_B)^{3/2}}{2\pi^2 h^3} \int_0^\infty \frac{y^2}{e^{y^2} - 1} dy$$

$$N = \frac{V(2mK_B T_B)^{3/2}}{2\pi^2 h^3} \left(2 \cdot 612 \frac{\sqrt{\pi}}{4} \right)$$

$$T_B = \left[\frac{N}{2 \cdot 612 V} \right]^{2/3} \frac{2\pi h^2}{m K}$$

$$T_B \propto \left(\frac{N}{V} \right)^{2/3}$$

$$T_B \propto n^{2/3}$$

$$T_B \propto \frac{1}{m}$$

→ what happens when T decreases beyond T_B
 $(T < T_B)$

There is a problem:

Problem is that zero weightage is given to the g.s.
 $\epsilon = 0$ as $g(\epsilon) \propto \epsilon^{1/2}$

$$\text{If } \epsilon = 0, \quad g(\epsilon=0) = 0$$



$$\text{But } \langle n \rangle = \frac{1}{e^{\beta(\epsilon - H)} - 1}$$

$$\text{if } \epsilon = 0 \text{ & } H = 0$$

$$\langle n \rangle = \infty$$

This problem can be solved by isolating the state s let
gr st has No particles.

Then

$$N = N_0 + \frac{V}{2\pi^2\hbar^3} \int_0^\infty \frac{p^2 dp}{(e^{p^2/2m\kappa_B T} - 1)} \rightarrow \frac{V(2m\kappa_B T)^{3/2}}{2\pi^2\hbar^3} \left(\frac{0.6125\pi}{T}\right)^{3/2}$$

$$N = N_0 + N \left(\frac{T}{T_B}\right)^{2/3}$$

$$N_0 = N \left[1 - \left(\frac{T}{T_B}\right)^{2/3} \right]$$



$$T < T_B$$

$T \rightarrow 0$ (population increases in gr. st.)
 $N_0 \rightarrow N$ (i.e., all particles goes over to the gr. state)

Energy of Bosonic system below T_B \rightarrow

$$\langle n \rangle = \frac{1}{e^{\beta(H)} - 1}$$

$$T < T_B$$

$E_n(E)$ of Bosonic system below T_B

$$E = \int g(E) dE \times f(E) \times E$$

$$= \frac{V}{2\pi^2\hbar^3} \int \frac{p^2}{2m} \cdot \frac{p^2 dp}{e^{\beta(p^2/2m)} - 1}$$

$$= \frac{V}{4\pi^4 h^3 m} (2m k_B T)^{5/2} \int_0^\infty \frac{y^4}{e^{y^2 - 1}} dy$$

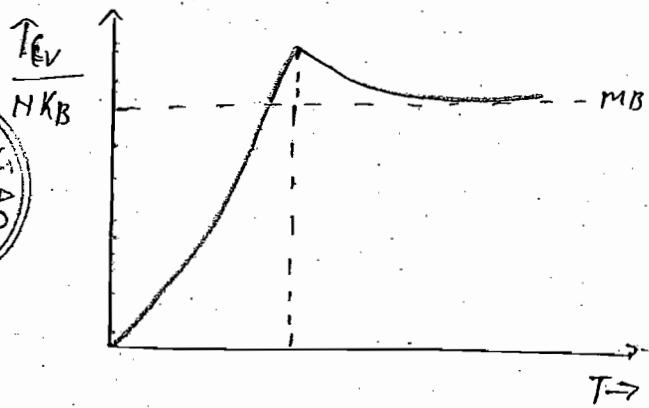
$$E = \frac{V}{4\pi^4 h^3 m} (2m k_B T)^{5/2} \frac{3}{8} \times 1.391 \sqrt{\pi}$$

$$E = 0.77 N k T^{5/2}$$



$$C_V = \frac{dE}{dT}$$

$$C_V \propto T^{3/2}$$



Q. Determine the Bose temp. of Bosons each of mass of $6.65 \times 10^{-27} \text{ kg}$ and spin 0, their concentration being $10^{26}/\text{m}^3$. Then also determine the fraction N_0/N of the Bosons in gr. st. at a temp. $0.2 T_B$

$$T_B = 1.32 \text{ K}, \quad \frac{N_0}{N} = 0.91$$

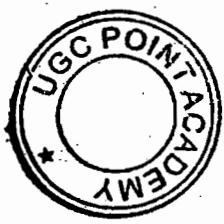
Q. The excitation of 3-D solid are bosonic in nature with their freq. ω & wave vector \mathbf{k} , which are related as $\omega \propto k^2$. In large wavelength limit the chemical pot. is zero the behaviour of sp. heat depends on T as

$$T, T^3, T^{3/2}, T^{5/2}$$

Ans: $E = \int E g(\epsilon) d\epsilon f(\epsilon)$

$$= \int e^{-\frac{4\pi p^2 dp}{h^3}} \frac{1}{e^{\beta E} - 1}$$

$$\propto \int e^{-\epsilon^{1/2}} \frac{1}{e^{\beta E} - 1}$$



$$\int_0^\infty \frac{\epsilon^{3/2}}{e^{\beta E} - 1} d\epsilon$$

$$\propto T^{5/2} \int_0^\infty \frac{x^{3/2} dx}{e^x - 1}$$

$$CV \propto T^{3/2}$$

→ Condensation of photons does not possible.

$$\langle n \rangle = \frac{1}{eBE - 1}$$

No. is not conserved in case of photon.

$E \propto p^5$
 $S=1$ condensation not possible.

Q. condensation is possible for

$$0 < S < 1$$

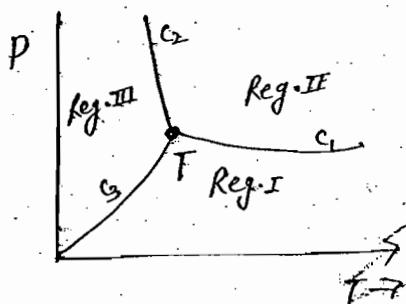
$$0 < S < 2$$

$$\begin{array}{c} S < 2 \\ 0 < S < 3 \\ 0 < S < 4 \end{array}$$



Phase transition :

Ex: Water - vapour



T → Triple pt.

Point where all the three phase, solid, liquid & gas are in equilibrium with each other.

In case of water the co-ordinate of triple pt.
($0, 075^\circ\text{C}$, 76cm)

Reg. I

Gas

Curve 1

Curve of
vapourisation

Reg. II

liquid

Curve 2

Curve of
fusion

Reg. III

solid

Curve 3

Curve of sublimation.

Critical point:

The p_c at which liquid vapour equilibrium lying ends.
Above critical point, there is no further phase transition.

Prob: The vapour pressure P (in mm of Hg) of a solid at temp. T is expressed by $\ln P = 23 - \frac{3863}{T}$ & that of liquid phase by $\ln P = 19 - \frac{3063}{T}$



Q. At transition p_c which is correct
 $G_f = 0$ one-state to other $\xrightarrow{T} \xrightarrow{P}$
 $dG_f = \text{constt}$
 $dG_f = 0$
 None

free En G_f i.e. G_f(T, P) is continuous at T_p.

$$\underline{dG_f = 0}$$

$$* G_f = H - TS = U + PV - TS$$

$$dG_f = \underline{dU + pdV + VdP} - VdS - SdT$$

$$dG_f = VdP - SdT$$

or

$$dG_f = VdP - SdT - MdH$$

Types of transitions

Ist order

βT derivative of
 G_f is discontinuous

Ex: V , S , M , Density
↓
volume, Entropy, magnetization

IInd order

2nd order derivative is discontinuous
& 1st order derivative is continuous

$C_v, \chi_m, \text{ compressibility}$

$$K_T = -\frac{1}{V} \left(\frac{\partial V}{\partial P} \right)$$

Ising model:

A model used to explain ferromagnetism easily.

$$E = -J \sum_{ij} s_i s_j - M H \sum_i s_i$$

↓
dipole moment of atom
mag. field

Interaction En. with nearest neighbouring atom
 s_i, s_j can take value ± 1 only.

$J \rightarrow$ exchange En.

If $H = 0$

$$E = -J \sum_{ij} s_i s_j$$

let there are 2 atom

$$E = -J \sum s_1 s_2 = -J \sum_{s_1} s_1 \sum_{s_2} s_2$$

$$E = e^{\beta J} \sum_{s_1, s_2 = \pm 1} s_1 s_2 \quad \text{or} \quad \sum_{s_1} \sum_{s_2} e^{\beta J} s_1 s_2 \quad s_1 s_2 = \pm 1$$

$$E = e^{\beta J} \sum_{s_1} s_1 (+1) + e^{\beta J} \sum_{s_1} s_1 (-1) = \sum_{s_1} e^{\beta J s_1 (+1)} + \sum_{s_1} e^{\beta J s_1 (-1)}$$

$E_2 = 2^2 (\cosh \beta J)^2$

If there are 3 atom

$$E_3 = 2^3 (\cosh \beta J)^2$$

In case of N-particle

$$Z_n = 2^N (\cosh \beta J)^{N-1}$$

$$F = -k_B T \ln Z_N$$

If $H \neq 0$

$$Z_N = 2^N (\cosh \beta J)^{N-1} (2 \cosh \beta H)^N$$



126

